

Section 6-1

Concept Review

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 10 neutrons, and 12 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a octet.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

noble gases are stable while everything
else is missing some

Concept Review Simple Ions

**Geoff Rayner-Canham, Geoffrey
Rayner-Canham, Tina Overton**



Concept Review Simple Ions:

Concepts in Biology' 2007 Ed.2007 Edition , *Hazmat Chemistry Study Guide (Second Edition)* Jill Meryl Levy,2011 Holt Chemistry Ralph Thomas Myers,2004 *Principles of Life* David M. Hillis,2012 For sample chapters a video interview with David Hillis and more information visit www.whfreeman.com/hillispreview Sinauer Associates and W H Freeman are proud to introduce Principles of Life Written in the spirit of the reform movement that is reinvigorating the introductory majors course Principles of Life cuts through the thicket of excessive detail and factual minutiae to focus on what matters most in the study of biology today Students explore the most essential biological ideas and information in the context of the field s defining experiments and are actively engaged in analyzing research data The result is a textbook that is hundreds of pages shorter and significantly less expensive than the current majors introductory books International Review of Neurobiology ,1963-01-01 International Review of Neurobiology **Columbia Law Review** ,1922 *Interatomic Potentials and Simulation of Lattice Defects* P. Gehlen,2012-12-06 This book is the proceedings of the Sixth Battelle Colloquium on the Science of Materials The Colloquium was devoted to a new field of materials science in which computers are used to conduct the experiments Although the computer methods used have reached a high degree of sophistication the underlying principles are relatively straightforward and well understood The interatomic force laws a vital input into these computations however are less well understood Interatomic Potentials and Simulation of Lattice Defects primarily discusses the validity of a variety of force laws either from a theoretical point of view or through comparisons of experimental results and those obtained with computer simulation The format used in previous Battelle Institute Colloquia is followed The opening session was aimed at providing an overall view of the field of interatomic forces and defect calculations by major contributors It was led by Dr G H Vineyard one of the pioneers in this field The second day was devoted to research papers on theoretical and experimental aspects of interatomic forces The remaining days were devoted to research papers on computer simulation of the four types of defects point defects line defects surface defects and volume defects **Ionic Reactions and Equilibria** Omer Robbins,1967 Descriptive Inorganic Chemistry Geoff Rayner-Canham,Geoffrey Rayner-Canham,Tina Overton,2009-12-28 This bestselling text introduces descriptive inorganic chemistry in a less rigorous less mathematical way The book uses the periodic table as basis for understanding chemical properties and uncovering relationships between elements in different groups Rayner Canham and Overton s text also familiarizes students with the historical background of inorganic chemistry as well as with its crucial applications especially in regard to industrial processes and environmental issues resulting in a comprehensive appreciation and understanding of the field and the role it will play in their fields of further study *Reviews of Pure and Applied Chemistry* ,1967

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