ä

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- the relative number of atoms in a molecule
   one formula mass of water
   one gram of water
- 6. A gas has a density of 0.717 g/l at STP. What is the mass of one mole of this gas?
  - a. 8.0 g c. 28.1 g b. 16.1 g d. 4.0 g
- The molecular mass of CO<sub>2</sub> is 44; therefore, at STP a. 1 liter of CO<sub>2</sub> weighs 44 g
   b. 11.2 liters of CO<sub>2</sub> weigh 22 g
   c. the molar volume of CO<sub>2</sub> is 44
  - d. 22.4 g. of CO<sub>2</sub> have a volume of 44 liters
- The formula, which gives the simple ratio of each kind of atoms present in the molecule of compound, is called
  - a. Molecular Formula
    b. Empirical Formula
    c. Structural Formula
    d. formula unit
- Smallest particle of an element which may or may not have independent existence is known as.
  - a. A molecule
     b. An ion
     c. An atom
     d. An electron
- Atoms and molecules can either gain or lose electrons, forming charged particles called:
   a. Positrons
   b. Atoms

# **Chemistry Stoichiometry Mcqs**

**Arshad Iqbal** 

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