## Acids and Bases

How do acids and bases behave in water?

# Why?

Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water. In this activity, we will explore the physical and chemical properties of acids and bases.

#### Model 1 – Arrhenius Acids and Bases

Common Name for Aqueous Solution	Chemical Formula	Found in	Tastes	Turns Litmus Paper	Conducts Electricity?	Acid or Base
Acetic acid	HC,H,O,(aq)	Vinegar	Sour	Red	Yes	Acid
Benzoic acid	HC,H,O,(aq)	Food preservative	Sour	Red	Yes	Acid
Phosphoric acid	H,PO4(aq)	Soda pop	Sour	Red	Yes	Acid
Hydrochloric acid	HCl( aq)	Stomach acid	Sour	Red	Yes	Acid
Citric acid	H <sub>4</sub> C <sub>4</sub> H <sub>4</sub> O <sub>5</sub> (aq)	Citrus fruits	Sour	Red	Yes	Acid
Ascorbic acid	$H_2C_aH_aO_a(aq)$	Vitamin C	Sour	Red	Yes	Acid
Magnesium hydroxide	Mg(OH),(aq)	Milk of magnesia	Bitter	Blue	Yes	Base
Aluminum hydroxide	Al(OH),(aq)	Antacids	Bitter	Blue	Yes	Basse
Barium hydroxide	Ba(OH),(aq)	Lubricants	POISON	Blue	Yes	Base
Sodium hydroxide	NaOH(aq)	Drain cleaner	POISON	Blue	Yes	Base

- 1. Refer to Model 1.
  - What is the common chemical name for Vitamin C? ascorbic acid
  - b. Is Vitamin C classified as an acid or a base? acid
- Examine the properties of the Arrhenius acids in Model 1. List three properties that all Arrhenius acids have in common.

sour, red, conduct electricity

- 3. Examine the chemical formulas for the Arrhenius acids in Model 1. What feature do all the Arrhenius acid chemical formulas have in common?
- Examine the properties of the Arrhenius bases in Model 1. List two properties that all Arrhenius bases have in common.

# **Chemistry Pogil Answers Acids And Bases**

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Organic Chemistry Suzanne M. Ruder, The POGIL Project, 2015-12-29 ORGANIC CHEMISTRY Analytical Chemistry Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the BerriLambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more

Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes Chemistry Education Javier García-Martínez, Elena Serrano-Torregrosa, 2015-02-17 Winner of the CHOICE Outstanding Academic Title 2017 Award This comprehensive collection of top level contributions provides a thorough review of the vibrant field of chemistry education Highly experienced chemistry professors and education experts cover the latest developments in chemistry learning and teaching as well as the pivotal role of chemistry for shaping a more sustainable future Adopting a practice oriented approach the current challenges and opportunities posed by chemistry education are critically discussed highlighting the pitfalls that can occur in teaching chemistry and how to circumvent them The main topics discussed include best practices project based education blended learning and the role of technology including e learning and science visualization Hands on recommendations on how to optimally implement innovative strategies of teaching chemistry at university and high school levels make this book an essential resource for anybody interested in either teaching or learning chemistry more effectively from experience chemistry professors to secondary school teachers from educators with no formal training in didactics to frustrated chemistry students Overcoming Students' Misconceptions in Science Mageswary Karpudewan, Ahmad Nurulazam Md Zain, A.L. Chandrasegaran, 2017-02-28 This book discusses the importance of identifying and addressing misconceptions for the successful teaching and learning of science across all levels of science education from elementary school to high school It suggests teaching approaches based on research data to address students common misconceptions Detailed descriptions of how these instructional approaches can be incorporated into teaching and learning science are also included The science education literature extensively documents the findings of studies about students misconceptions or alternative conceptions about various science concepts Furthermore some of the studies involve systematic approaches to not only creating but also implementing instructional programs to reduce the incidence of these misconceptions among high school science students These studies however are largely unavailable to classroom practitioners partly because they are usually found in various science education journals that teachers have no time to refer to or are not readily available to them In response this book

offers an essential and easily accessible guide **Acids and Bases** Lynnette Brent, 2008-08-15 Did you know that cola is an acid And your saliva is a base Young readers will learn about common acids and bases from lemon juice to ammonia Through vivid examples and exciting illustrations this book will eagerly explore these important chemical compounds **Bases** Kristi Lew, 2009 Learn about acids and bases chemical components of the natural world that play key roles in medicine Chemists' Guide to Effective Teaching Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe, 2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry and Solutions Elizabeth R. Kellerman, 1999-09-01 The activities in this book explain elementary concepts in the study of chemistry including acids bases solvents solutions crystals and more General background information suggested activities questions for discussion and answers are included Acids and Bases Daniel R. Faust, 2023 Why does a baking soda and vinegar volcano erupt That's what happens when you mix and an acid and a base But just what are acids and bases What makes them so different Learn the answers to these questions and more It's key chemistry curriculum made approachable for all

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## **Table of Contents Chemistry Pogil Answers Acids And Bases**

- 1. Understanding the eBook Chemistry Pogil Answers Acids And Bases
  - The Rise of Digital Reading Chemistry Pogil Answers Acids And Bases
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Chemistry Pogil Answers Acids And Bases
  - Exploring Different Genres
  - $\circ\,$  Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Pogil Answers Acids And Bases
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chemistry Pogil Answers Acids And Bases
  - Personalized Recommendations
  - Chemistry Pogil Answers Acids And Bases User Reviews and Ratings
  - Chemistry Pogil Answers Acids And Bases and Bestseller Lists
- 5. Accessing Chemistry Pogil Answers Acids And Bases Free and Paid eBooks
  - Chemistry Poqil Answers Acids And Bases Public Domain eBooks
  - Chemistry Pogil Answers Acids And Bases eBook Subscription Services
  - Chemistry Pogil Answers Acids And Bases Budget-Friendly Options
- 6. Navigating Chemistry Pogil Answers Acids And Bases eBook Formats
  - o ePub, PDF, MOBI, and More
  - Chemistry Pogil Answers Acids And Bases Compatibility with Devices
  - Chemistry Pogil Answers Acids And Bases Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Pogil Answers Acids And Bases
  - Highlighting and Note-Taking Chemistry Pogil Answers Acids And Bases
  - Interactive Elements Chemistry Pogil Answers Acids And Bases
- 8. Staying Engaged with Chemistry Pogil Answers Acids And Bases

- o Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Chemistry Pogil Answers Acids And Bases
- 9. Balancing eBooks and Physical Books Chemistry Pogil Answers Acids And Bases
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Pogil Answers Acids And Bases
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Pogil Answers Acids And Bases
  - Setting Reading Goals Chemistry Pogil Answers Acids And Bases
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Pogil Answers Acids And Bases
  - Fact-Checking eBook Content of Chemistry Pogil Answers Acids And Bases
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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