CHEMICAL REACTIONS & EQUATIONS

1. Photosynthesis Reaction:

$$6CO_2(aq) + 12H_2O(l) \xrightarrow{\text{Sunlight}} C_6H_{12}O_6(aq) + 6O_2(aq) + 6H_2O(l)$$
 (1.12)
(Glucose)

2. Formation of Slaked Lime

CaO(s) +
$$H_2O(l) \rightarrow Ca(OH)_2(aq)$$

(Quick lime) (Slaked lime)

(it is also called Ca(OH)2, so don't get confused)

Whitewashing of walls: Calcium hydroxide reacts slowly
with the carbon dioxide in air to form a thin layer of calcium
carbonate on the walls. Calcium carbonate is formed after
two to three days of white washing and gives a shiny finish
to the walls. It is interesting to note that the chemical
formula for marble is also CaCO3.

$$Ca(OH)_2(aq) + CO_2(g) \rightarrow CaCO_3(s) + H_2O(l)$$

(Calcium
hydroxide) (Calcium
carbonate)

4. Decomposition Reaction:

Chemical Equations And Reactions Study Guide Answers

Martha Mackin

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Chemistry Equations And Answers (Speedy Study Guides) Speedy Publishing, 2015-04-24 Chemistry is a difficult subject to fully comprehend with its equations and scientific laws Trying to digest an entire book in one semester is a tough job but with the help of study guides like these you can absorb information in chemistry much more effectively This guide covers chemical equations including examples potential problems and solutions The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter Study Guide to Accompany Calculus for the Management, Life, and Social Sciences Clyde Metz,1984-01-01 Study Guide to Accompany Calculus for the Management Life and Social Sciences Chemistry **Equations & Answers** Speedy Publishing, 2014-08-01 One of the most important parts of learning chemistry is simply knowing all of the equations and formulas that are used in it It is important to know the exact formatting of these equations and most classes will require a student to know them for exams It is a good idea to learn these equations and formulas with the use of a study pamphlet The pamphlet can condense all of the information so a student can memorize the equations and Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front formulas while studying cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to

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