

Chemistry 530 Midterm Review

Exam on Monday Dec 16 at 8:30

Section I = 75 pts

- Consists of 75 multiple choice questions worth 1 pt. each
- Questions cover all material covered during the second semester
- Questions will be similar to multiple choice problems in the book and review guides on-line
- Practicing the review guides is the best approach; answers will be posted on-line

Section II = 95 pts

- Consists of 19 short answer problems totaling 95 points
- Questions cover the whole semester
- Examples of types of questions
 - Metric Conversions
 - Volume Calculations
 - Percent Composition
 - Atomic Mass
 - Electron configuration
 - Nomenclature
 - Balancing Equations
 - Predicting the products of a reactions
 - Identifying types of reactions
 - Mole Conversions
 - Molar Mass of a compound
 - Molar Mass of a gas at STP
 - Density of a gas at STP
 - Empirical & Molecular Formula

Study approach

- First, read through your notes. Once thoroughly and once more quickly.
- Go through the review packet and make sure you are comfortable with all of the terms and how to do problems
- Do the review packets for help with multiple choice
- Do questions on old tests for practicing written problems
- Make sure you have practiced AT LEAST one of the types of example problems

Chapter 1 – Introduction to Chemistry

- Chemistry: the branch of science that studies the composition and properties of matter.
- Scientific Method:
 - Observation
 - Hypothesis
 - Controlled Experiment
- Theory versus Natural Law

Prerequisite Science Skills

- Rules for Significant Figures:
 - 1) All non-zero digits are significant.
 - 2) Zeros are significant only when they are *between* two non-zero digits or if they occur *after* a decimal point and *after* a non-zero digit.
 - 3) In *multiplication or division*, you cannot end up with more significant figures than you started with.
 - 4) In *addition or subtraction*, you cannot have more digits to the *right* of the decimal than you started with.
- Exponential numbers
- Scientific Notation = $D.DD \times 10^n$ where $1.00 \leq D.DD \leq 9.99$

Chemistry Honors Midterm Study Guide Answer

Cathy Duffy



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Christian Home Educators' Curriculum Manual Cathy Duffy, 1997-11 **Eye on Apply** Princeton Review (Firm), 2004 In six compelling narratives MTV's Real World meets high stakes college admissions as high school seniors share the diaries they wrote while trying to get into college Participant Observer William Foote Whyte, 2019-05-15 While it documents a remarkable career Participant Observer is also a personal chronicle in which William Foote Whyte reflects on his childhood his education his courageous struggles with polio and with the crises of family and academic life Beginning with the study of gangs in Boston's North End recorded in Street Corner Society Whyte listened to what working people had to say becoming a powerful voice for worker participation and workplace democracy His career is a model for the social sciences and his story should be read by any serious student of them **The Advisor, Teacher-course Evaluation** University of Illinois at Urbana-Champaign. Student Senate, 1968 *Catalogs of Courses* University of California, Berkeley, 1993 Includes general and summer catalogs issued between 1878 1879 and 1995 1997 *Who's Who in the World*, 1995 Marquis Who's Who, 1995-12 This single volume affords instant access to more than 35 000 individual biographies of the people whose activities are shaping today's world Among those profiled are prominent government figures high ranking military officers leaders of the largest corporations in each country heads of religious organizations pioneers in science the arts many more **Study Guide for Chemical Principles [by] Steven S. Zumdahl** Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section's critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text Fast Track: Chemistry The Princeton Review, 2020-12-08 GET UP TO SPEED WITH FAST TRACK CHEMISTRY Covering the most important material taught in high school chem class this essential review book breaks need to know content into accessible easily understood lessons Inside this book you'll find Clear concise summaries of the most important concepts terms and functions in chemistry Diagrams charts and graphs for quick visual reference Easy to follow content organization and illustrations With its friendly straightforward approach and a clean modern design crafted to appeal to visual learners this guidebook is perfect for catching up in class or getting ahead on exam review Topics covered in Fast Track Chemistry include Atomic structure Covalent bonding Intermolecular forces Stoichiometry Precipitation reactions Gas laws Thermochemistry Equilibrium and the solubility product constant Redox reactions Electrochemistry Acids and bases Kinetics

and more General Chemistry Darrell D. Ebbing, Larry K. Krannich, Joan I. Senyk, 1996

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