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Chemical Equilibrium occurs when the concentrations of the chemical species remain the same with time having reached Equilibrium. Chemical reactions are represented by the reaction: $aA + bB \rightleftharpoons cC + dD$

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General Representation Chemical Equilibrium occurs when the concentrations of the chemical species remain the same with time having reached Equilibrium. Chemical reactions are represented by the reaction: $aA + bB \rightleftharpoons cC + dD$ Equilibrium Constant K_c $K_c = \frac{[C]^c [D]^d}{[A]^a [B]^b}$	Equilibrium Constant Equilibrium Constant K_c $K_c = \frac{[C]^c [D]^d}{[A]^a [B]^b}$ Equilibrium Constant K_p $K_p = \frac{P_C^c P_D^d}{P_A^a P_B^b}$	Graphical Representation of Equilibrium
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Le Chatelier's Principle: the effect of changing conditions on equilibrium		
Concentration 1. Adding more reactants shifts the equilibrium to the right. 2. Removing more reactants shifts the equilibrium to the left.	Pressure 1. Increasing the pressure shifts the equilibrium to the side with fewer moles of gas. 2. Decreasing the pressure shifts the equilibrium to the side with more moles of gas.	Temperature 1. Increasing the temperature shifts the equilibrium to the side that is endothermic. 2. Decreasing the temperature shifts the equilibrium to the side that is exothermic.

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**Barron's Educational Series, Mark
Kernion, Joseph A. Mascetta**



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far less emphasis on the skills needed to actually solve problems **Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today** Seager/Slabaugh,2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application *Organic Chemistry, 5e Student Study Guide and Solutions Manual* David R. Klein,Laurie S.

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