

BALANCING CHEMICAL EQUATIONS

Load the simulation Balancing Chemical Equations This link is in your JupiterGrades messages http://phet.colorado.edu/en/simulation/balancing-chemical-equations

CRITICAL THINKING QUESTIONS

REVIEW:

MOLECULES: A GROUP OF ATOMS BONDED TOGETHER ATOMS: THE BASIC UNIT OF A CHEMICAL ELEMENT

- 1. Explore the Balancing Chemical Equations simulation. Discuss with your group what you find.
 - a) What are the tools you can use in the simulation to tell you when an equation is balanced? Scale or bar graph
- 2. For each balanced reaction, indicate the total number of molecules in the table below.

Reaction	Total Number of Molecules	
	Reactant Side (Left)	Product Side (Right)
Make Ammonia	나	2
Separate Water	2	3
Combust Methane	3	3

3. Is the number of total molecules on the left side of a balanced equation always equal to the number of total molecules on the right side of the equation? Explain your answer.

No because the atoms rearrange in a chemical reaction to form new molecules.

For each balanced reaction, indicate the total number of atoms in the table below.

Reaction	Total Number of Atoms	
	Reactant Side (Left)	Product Side (Right)
Make Ammonia	Q	8
Separate Water	6	ما
Combust Methane	9	q

5. Is the number of total atoms on the left side of a balanced equation always equal to the number of total atoms on the right side of the equation?

Yes

- What is the same on the left and right side of a balanced equation? Explain your answer. The numbers are all the same because there balanced.
- As a group, play level 1 of the balancing equation game. Write down the strategies your group uses to balance chemical equations.

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