#### Chapter 2 Chemistry Assessment Answers

#### 2.1 Assessment Answers

- 1. Intensive and extensive properties
- 2. Every sample of a given substance has the same chemical composition.
- 3. Solid, liquid, gas
- Physical changes are either reversible or irreversible. Reversible changes can be "undone", or reversed. Irreversible changes cannot be undone.
- Color; sodium chloride is the only white solid listed.
- Liquids and gases have indefinite shape. The shape of a solid is definite; the shape of a liquid is indefinite.
- 7. The freezing of mercury is reversible because solid mercury can be melted.
- 8. Samples of gold and copper can have the same mass and volume "extensive properties).

They cannot have the same set of intensive properties because they have different chemical compositions.

#### 2.2 Assessment Answers

- 11. As heterogeneous or homogeneous
- 12. Differences in physical properties.
- 13. A phase is any part of a sample with uniform composition. There is one phase in a homogeneous mixture and two or more phases in a heterogeneous mixture.
- 14.
- a. homogeneous
- b. heterogeneous
- c. homogeneous
- d. heterogeneous
- Both have a uniform composition throughout. A substance has a definite coimposotion; a solution has a variable composition.
- Filtrations separates solids from a liquid in a heterogeneous mixture. Distillation can separate a liquid from a substance dissolved in the liquid.
- 17. Add water to dissolve the salt. Pour the resulting mixture onto a piece of closely woven cloth. The sand will remain on the cloth, and the salt solution will pass through. Use evaporation to remove the water from the salt solution, leaving solid salt behind.

# **Chemistry Chapter 61 Assessment Answers**

**Disha Experts** 

#### **Chemistry Chapter 61 Assessment Answers:**

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## **Table of Contents Chemistry Chapter 61 Assessment Answers**

- 1. Understanding the eBook Chemistry Chapter 61 Assessment Answers
  - The Rise of Digital Reading Chemistry Chapter 61 Assessment Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Chemistry Chapter 61 Assessment Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Chapter 61 Assessment Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chemistry Chapter 61 Assessment Answers
  - Personalized Recommendations
  - Chemistry Chapter 61 Assessment Answers User Reviews and Ratings
  - Chemistry Chapter 61 Assessment Answers and Bestseller Lists
- 5. Accessing Chemistry Chapter 61 Assessment Answers Free and Paid eBooks
  - Chemistry Chapter 61 Assessment Answers Public Domain eBooks
  - Chemistry Chapter 61 Assessment Answers eBook Subscription Services
  - Chemistry Chapter 61 Assessment Answers Budget-Friendly Options
- 6. Navigating Chemistry Chapter 61 Assessment Answers eBook Formats

- o ePub, PDF, MOBI, and More
- Chemistry Chapter 61 Assessment Answers Compatibility with Devices
- Chemistry Chapter 61 Assessment Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Chapter 61 Assessment Answers
  - Highlighting and Note-Taking Chemistry Chapter 61 Assessment Answers
  - Interactive Elements Chemistry Chapter 61 Assessment Answers
- 8. Staying Engaged with Chemistry Chapter 61 Assessment Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - $\circ$  Following Authors and Publishers Chemistry Chapter 61 Assessment Answers
- 9. Balancing eBooks and Physical Books Chemistry Chapter 61 Assessment Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Chapter 61 Assessment Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Chapter 61 Assessment Answers
  - $\circ\,$  Setting Reading Goals Chemistry Chapter 61 Assessment Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Chapter 61 Assessment Answers
  - Fact-Checking eBook Content of Chemistry Chapter 61 Assessment Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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