

Combined Gas Law Problems

- 1) A sample of sulfur dioxide occupies a volume of 652 mL at $40.^{\circ}\text{C}$ and 720 mm Hg. What volume will the sulfur dioxide occupy at STP?
- 2) A sample of argon has a volume of 5.0 dm^3 and the pressure is 0.92 atm. If the final temperature is $30.^{\circ}\text{C}$, the final volume is 5.7 L, and the final pressure is 800. mm Hg, what was the initial temperature of the argon?
- 3) 322 L of hydrogen occupies a volume of 197 L at STP. If the initial temperature of the hydrogen was 37°C , what was its initial pressure?
- 4) The initial temperature of a 1.00 liter sample of argon is $20.^{\circ}\text{C}$. The pressure is decreased from 720 mm Hg to 360 mm Hg and the volume increases to 2.14 liters. What was the change in temperature of the argon?
- 5) A sample of nitrogen gas occupies a volume of 2.00 L at 756 mm Hg and 0.00°C . The volume increases by 2.00 L and the temperature decreases to 137 K. What is the final pressure exerted on the gas?
- 6) A 20. L container is filled with helium and the pressure is 150 atm and the temperature is $30.^{\circ}\text{C}$. How many 5.0 L balloons can be filled when the temperature is 22°C and the atmospheric pressure is 755 mm?

Combined Gas Law Study Guide Answers

**John Krenos, Joseph Potenza, Carl
Hoeger**



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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises *Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry* Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text ***Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual*** Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter

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The Complete Chemistry Study Guide and Note Cards and MCAT Konstantinos Papadopoulos, 2012-07-06

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