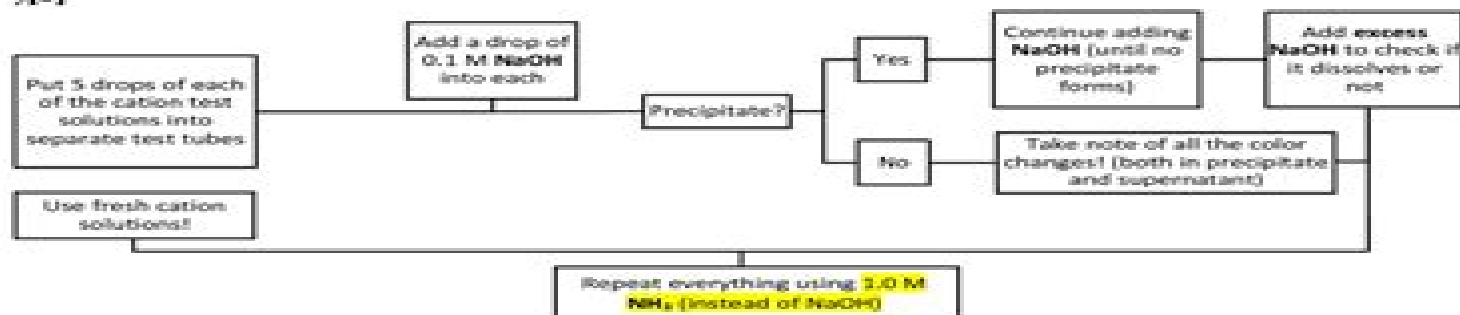


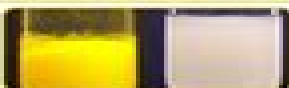
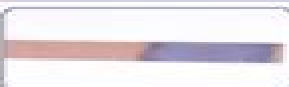
1. CATIONS – Ca^{2+} , Fe^{3+} , Cu^{2+} , Zn^{2+} , NH_4^+

Elimination Tests

A-1



Confirmatory Tests

	Cu^{2+} • Deep blue complex (upon mixing of solution with excess NH_3)
	Fe^{3+} • Blood red complex (FeSCN^{2+}) (upon addition of 2 drops of 0.1 M KSCN) • Dark blue precipitate ($\text{KFe}^{III}\text{Fe}^{II}(\text{CN})_6$) (upon addition of 2 drops of 0.1 M $\text{K}_4\text{Fe}(\text{CN})_6$)
	Ca^{2+} • White precipitate (upon mixing with 2 drops of 0.1 M $(\text{NH}_4)_2\text{C}_2\text{O}_4$)
	Zn^{2+} • Yellow precipitate ($\text{K}_2\text{ZnFe}(\text{CN})_6$) (upon addition of 2 drops of 0.1 M $\text{K}_4\text{Fe}(\text{CN})_6$) • White precipitate ($\text{K}_2\text{Zn}(\text{C}_2\text{O}_4)_2 \cdot 2\text{H}_2\text{O}$) (upon addition of 0.1 M $\text{K}_2\text{C}_2\text{O}_4$)
	NH_4^+ • Red litmus paper turns blue upon contact with vapor formed by mixing 30 drops of solution with 5 drops of 1.0 M NaOH



Cations And Ions Lab Report

Puja Mehta



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