

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

**CHAPTER 9** STUDY GUIDE FOR CONTENT MASTERY

## Covalent Bonding

### Section 9.1 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

|               |          |            |            |        |
|---------------|----------|------------|------------|--------|
| covalent bond | molecule | sigma bond | exothermic | p bond |
|---------------|----------|------------|------------|--------|

When sharing of electrons occurs, the attachment between atoms that results is called a(n) **(1) covalent bond**. When such an attachment is formed, bond dissociation energy is released, and the process is **(2) exothermic**. When two or more atoms bond by means of electron sharing, the resulting particle is called a(n) **(3) molecule**. If the electrons shared are contained between the two atoms, the attachment is called a(n) **(4) sigma bond**. If the sharing involves the overlap of parallel orbitals, the attachment is called a(n) **(5) p bond**.

In your textbook, read about single and multiple bonds and bond strengths.

Circle the letter of the choice that best completes the statement or answers the question.

- In what form do elements such as hydrogen, nitrogen, and oxygen normally occur?
  - a single atom
  - a molecule containing three atoms**
  - a molecule containing two atoms
  - a molecule containing four atoms
- How many electrons are shared in a double covalent bond?
  - none
  - one
  - two
  - four**
- Bond lengths in the distance between
  - two molecules of the same substance
  - the nuclei of two attached atoms**
  - the electrons in two attached atoms
  - the orbitals of two attached atoms
- Which of the following relationships relating to bond length is generally correct?
  - the shorter the bond, the stronger the bond**
  - the shorter the bond, the weaker the bond
  - the shorter the bond, the fewer the electrons in it
  - the shorter the bond, the lower the bond dissociation energy

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Chemistry: Matter and Change • Chapter 9

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**CHAPTER 9** STUDY GUIDE FOR CONTENT MASTERY

### Section 9.2 Naming Molecules

In your textbook, read about how binary compounds and acids are named from their formulas.

For each statement below, write *true* or *false*.

- false** Binary molecular compounds are generally composed of a metal and a nonmetal.
- true** The second element in the formula of a binary compound is named using the suffix *-ite*.
- false** The prefix *tri-* indicates three atoms.
- true** The prefix *kilo-* indicates six atoms.
- true** In naming the first element in a formula, the prefix *mono-* is not used.
- true** For binary acids, the hydrogen part of the compound is named using the prefix *hydro-*.
- false** An crystal contains only two elements.
- false** If the name of the cation of an ionic compound is only the acid name contains the suffix *-ate*.

In your textbook, read about naming molecular compounds and oxoacids.

For each item in Column A, write the letter of the matching item in Column B.

| Column A                   | Column B             |
|----------------------------|----------------------|
| c. $\text{CO}$             | a. hydrobromic acid  |
| f. $\text{CO}$             | b. nitrogen trioxide |
| g. $\text{H}_2\text{CO}_3$ | c. carbon monoxide   |
| h. $\text{NO}_2$           | d. nitrous acid      |
| b. $\text{N}_2\text{O}_5$  | e. acetone           |
| d. $\text{HNO}_3$          | f. nitric acid       |
| i. $\text{HNO}_3$          | g. carbonic acid     |
| a. $\text{HBr}$            | h. bromic acid       |
| h. $\text{HBrO}_3$         | i. carbon dioxide    |

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Chemistry: Matter and Change • Chapter 9

# Chemistry Study Guide Content Mastery Teacher

**Guillaume Favre**



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*Handbook of Reading Research, Volume III* Michael L. Kamil,Peter B. Mosenthal,P. David Pearson,Rebecca Barr,2016-11-18 In Volume III as in Volumes I and II the classic topics of reading are included from vocabulary and comprehension to reading instruction in the classroom and in addition each contributor was asked to include a brief history that chronicles the legacies within each of the volume s many topics However on the whole Volume III is not about tradition Rather it explores the verges of reading research between the time Volume II was published in 1991 and the research conducted after this date The editors identified two broad themes as representing the myriad of verges that have emerged since Volumes I and II were published 1 broadening the definition of reading and 2 broadening the reading research program The particulars of these new themes and topics are addressed      [Encyclopedia of Educational Leadership and Administration](#) Fenwick W. English,2006-02-16 To read some sample entries or to view the Readers Guide click on Sample Chapters Additional Materials in the left column under About This Book The Encyclopedia of Educational Leadership and Administration presents the most recent theories research terms concepts ideas and histories on educational leadership and school administration as taught in preparation programs and practiced in schools and colleges today With more than 600 entries written by more than 200 professors graduate students practitioners and association officials the two volumes of this encyclopedia represent the most comprehensive knowledge base of educational leadership and school administration that has as yet been compiled Key Features Represents a knowledge dynamic of the field by presenting ideas and perspectives that are in the minds hearts and aspirations of those practicing in the profession Includes a wide range of topics covering teaching and learning curriculum psychology and motivation budgeting and finance law statistics research personnel management planning supervision and much more Contains more than 75 biographical sketches of people whose ideas aspirations and lives have contributed much to the profession Animates the reader s thinking and defines possibilities by presenting terms ideas concepts research and theories that are circulating in the field The Encyclopedia of Educational Leadership and Administration is a must have reference for all academic libraries as well as a welcome addition to any leadership in education collection      *High School Chemistry Teachers Magazine* ,1973      **Including the Gifted and Talented** Chris Smith,2006 Publisher description      **Proceedings of the International Conference on Social Science**

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