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The fundamental ideas in chemistry

electron
 nucleus
 neutron
 each for 1 mark

2.

(m)

- B D
- (ii) D
 (b) A and B only one type of atom
 C and D more than one type of atom
 accept element for atom
 ignore the word 'mixaire'
 (chemically) bonded

accept (chemically) joined or similar idea of joined

- 3. B
 carbon monoxide
 1

 CO
 accept carbon oxide
 1

 do not credit carbon dioxide
 1
 - do not credit if any superscripts or subscripts used but accept C₁O₁, accept OC do not credit if obviously lower case
 - C water
 H₂O accept hydrogen oxide
 do not accept hydrogen hydroxide
 do not credit if obviously lower case or if 2 not subscript
 do not accept HOH
 accept OHs
 - D ammonia do not accept ammonium 1
 NH3 do not credit if obviously lower case, or if 3 not subscript 1
 accept nitrogen hydride or hydrogen nitride
 do not accept hydrogen nitrate or nitrite
 allow H₃N
- 4. correct reactants (i.e. sodium + water either way round)
 - correct products (i.e. sodium hydroxide + hydrogen, either way round)
- correct use of 'react'/'reaction'/reactants'/combine (not mixed'added/join) 3
 correct use of 'produces'/products'/gives/forms/makes/creates reactants
 correctly identified

 each for 1 mark
 - products correctly identified



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Los Angeles Times ,1974

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