
Diffusion in Polymers 2020

The Diffusion Equation can model:

Absorption of solvent

Desorption of solvent

Permeation of solvent

Charles M. Hansen

Diffusion In Polymers Diffusion In Polymers

P. Neogi



Diffusion In Polymers Diffusion In Polymers:

Diffusion in Polymers P. Neogi, 1996-02-06 Examines various aspects of diffusion in polymers that are being quantitatively described and engineered detailing the phenomenology of diffusion and outlining areas for future research Emphasizing the importance of fundamental studies throughout *Diffusion in and Through Polymers* Wolf R. Vieth, 1991-01-01 Diffusion in Polymers, 1986 **Diffusion of Polymers in Polyelectrolyte Gels** Anand Rahalkar, 2016 Polyelectrolyte gels have the ability to undergo volume transitions in response to external stimuli As a result they are used in a wide variety of applications such as sensors actuators artificial implants and controlled drug delivery In addition diffusion of trapped macromolecules inside gels is of tremendous significance in drug delivery systems In this thesis the dynamics of trapped polyelectrolytes in a polyelectrolyte gel is studied using dynamic light scattering technique The effect of the crosslink density and the charge density of the gel as well as of solvent quality and salt concentration are investigated In general two distinct diffusive modes fast and slow are observed in light scattering experiments We have discovered that the fast mode corresponds to the gel elasticity and the slow mode corresponds to diffusion of trapped polymer The deduced diffusion coefficient of the trapped polymer obeys an exponential dependence on the volume fraction of the gel in accordance with the presence of an entropic barrier for polymer diffusion In addition we have monitored the diffusion of a neutral polymer in polyelectrolyte gels using fluorescence correlation spectroscopy We demonstrate that the polymer diffusion is considerably slowed down by the gel due to the coupling between polymer dynamics and gel dynamics **Polymer Permeability** J. Comyn, 2012-12-06 Polymers are permeable whilst ceramics glasses and metals are generally impermeable This may seem a disadvantage in that polymeric containers may allow loss or contamination of their contents and aggressive substances such as water will diffuse into polymeric structures such as adhesive joints or fibre reinforced composites and cause weakening However in some cases permeability is an advantage and one particular area where this is so is in the use of polymers in drug delivery systems Also without permeable polymers we would not enjoy the wide range of dyed fabrics used in clothing and furnishing The fundamental reason for the permeability of polymers is their relatively high level of molecular motion a factor which also leads to their high levels of creep in comparison with ceramics glasses and metals The aim of this volume is to examine some timely applied aspects of polymer permeability In the first chapter basic issues in the mathematics of diffusion are introduced and this is followed by two chapters where the fundamental aspects of diffusion in polymers are presented The following chapters then each examine some area of applied science where permeability is a key issue Each chapter is reasonably self contained and intended to be informative without frequent outside reference This inevitably leads to some repetition but it is hoped that this is not excessive *Third International Conference on Diffusion in Polymers*, 1991 *A Study of Diffusion in Polymers Using C-14 Labelled Molecules* H. K. Toh, 1970 A novel method of measuring the self diffusion coefficient of a penetrant molecule in various polymers was devised This method makes use of a permeation experiment

where the radioactively labelled molecules exchange with the chemically identical but nonradioactive molecules through the polymer membrane specimen. The rate of permeation is measured by the excitation of the non radioactive molecules which act as the solvent in a liquid scintillation mixture with subsequent excitation and fluorescence of the dissolved scintillator solutes. By maintaining the concentrations of the radioactive and non radioactive molecules in the vapour phase at the same level a self diffusion coefficient D at one precise penetrant concentration level can be determined. The diffusion coefficients measured were compared to those obtained from conventional sorption desorption methods and the comparison was discussed in terms of the basic definitions of the different diffusion coefficients. The diffusional behaviour in silicone rubber and S B S block copolymer was discussed mainly with reference to the free volume theory and the activated zone theory with particular emphasis on chain mobility. Dynamic mechanical studies were also obtained in these polymers and related to the diffusion characteristics through chain mobility and free volume concepts. Diffusion in filled systems and the two phase S B S copolymers was discussed with the help of mathematical models derived for analogous electrical conductivity through heterogeneous medium and a certain order of polystyrene domain distribution was indicated in the latter. Silica silicone rubber interaction was considered and some conclusions were made.

Polymer Properties Group Research Meeting X
Plastics and Rubber Institute. Polymers Properties Group. Research Meeting, 1981* *Diffusion in polymers*, ed John Crank, Geoffrey Sheard Park, *Diffusion in Polymers* James A. Tshudy, 1974

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