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E Durkheim



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examples are given in table 2 6 names of some ionic compounds table 2 6 compounds containing

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web bismuth has a molar mass of one mole of the compound would result in two moles of bismuth a total of 417 96g tellurium

has a molar mass of one mole of the compound would result in three moles of tellurium a total of 382 8g add the mass of

bismuth and the mass of tellurium per mole to find the total molar mass divide the mass of bismuth

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web in molecular compounds the atom binds each other through covalent bonds in salts it is held together with ionic bonds

these are the two types of bonds out of which every compound is made of 2 example of compounds example of compounds

includes water h₂o hydrogen peroxide h₂o₂ etc

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web the substances mentioned above exemplify the two basic types of chemical compounds molecular and ionic methane and

water are composed of molecules that is they are molecular compounds sodium chloride on the other hand contains ions it is

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web molecular compounds many compounds do not contain ions but instead consist solely of discrete neutral molecules these

molecular compounds covalent compounds result when atoms share rather than transfer gain or lose electrons covalent

bonding is an important and extensive concept in chemistry and it will be treated in considerable

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it is ionic write the symbols for the ions involved a nf 3 b ba o c nh 4 2 co 3 d sr h 2 po 4 2 e ibr f na 2 o answer a answer b

answer c answer d answer e answer f problem 4 3 1 4 4 3 1 4

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web may 28 2020 what is the total mass of hydrogen in each of the molecules a ch₄ b chcl₃ c c₁₂h₁₀o₆ d ch₃ch₂ch₂ch₃ answer a answer b answer c answer d problem 4 4 1 3 calculate the molecular or formula mass of each of the following a p₄ b h₂o c ca no₃ d ch₃co₂h acetic acid

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web hydrogen will be left slightly positive so this is a polar molecule i represent this as a couple of different animals here got a bear and it s got the five valence electrons and it shares the single electrons with its penguin friend the penguins each have one valence electron to be shared with from the bear

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web each dipole represents one polar bond the molecule is nonpolar 10 describe what you think is happening to the penguins in the illustration of h₂o on page 4 they are being pulled towards the stronger polar bear he s more electronegative 11 explain what you think the crossed arrow represents in the comic book polar bonds perhaps 12

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web the two polar bears or penguins are equal in electronegativity therefore they pull electrons equally as hard and neither one will end up with more electrons more of the time 4 a what 3 types of bonds are represented on the third page of the comic book nonpolar covalent polar covalent and ionic b

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web why does the polar bear tell the penguin to try to be more positive the polar bear benefits more when the penguin is more positive the more the polar bear attracts the shared pair of electrons the more positive the partial charge on the penguin

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what do you think the picture is trying to illustrate would hcl be attracted to the charged wand

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web to answer this question you will examine 1 electronegativity 2 nonpolar vs polar molecules 3 electronegativity and bonding electronegativity the h atom and the cl atom in hydrogen chloride

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web 1 how does the comic strip define a polar molecule 2 define electronegativity as you understand it after reading the first two pages of the comic strip 3 what is the artist trying to represent by two polar bears arm wrestling or two penguins arm wrestling 4 what three types of bonds are represented on the third page of the comic strip

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web explain how the iceberg penguins and polar bears represent trends in electronegativity on the periodic table and more study with quizlet and memorize flashcards containing terms like how does the comic define a polar molecule define electronegativity as you understand it after reading the first two pages of the comic interpret the pic

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web so the final representation of the nh₃ molecule using polar bears and penguins is as follows begin array c text polar bear nitrogen text with two lone pairs of electrons text and three penguins hydrogen end array this is a fun and creative way to represent the nh₃ molecule

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web polar bears and penguins purpose in this lesson you will be exploring polarity and bonding between atoms in greater detail a comic book will provide new information about these topics and will introduce you to the concept of electronegativity which helps us understand partial charges