

**Section 1.1** *continued*

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

|          |
|----------|
| 19       |
| 18       |
| Hydrogen |
| 1.008    |
| Helium   |

|                        |      |
|------------------------|------|
| Atomic Mass            | 1.01 |
| Atomic Number          | 18   |
| Electron Configuration | 18   |
| Chemical Name          | 18   |
| Chemical Symbol        | 18   |

Put each item in Column A, either the letter of the matching item in Column B.

**Column A**

- \_\_\_\_\_ 19. It exists in the periodic table.
- \_\_\_\_\_ 20. It has no the periodic table.
- \_\_\_\_\_ 21. Group 1 elements.
- \_\_\_\_\_ 22. Elements that are shiny and conduct electricity.
- \_\_\_\_\_ 23. Group 18 elements.

**Column B**

- a. metals
- b. group
- c. period
- d. representative elements
- e. noble gases

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- \_\_\_\_\_ 24. There are two representative elements of elements.
- \_\_\_\_\_ 25. Most of the elements in the periodic table are metals.
- \_\_\_\_\_ 26. Group 1 elements except for hydrogen are metals in the solid state.
- \_\_\_\_\_ 27. Group 18 elements are the noble gases.
- \_\_\_\_\_ 28. Group 1 elements are highly reactive elements in nature.
- \_\_\_\_\_ 29. Group 18 elements are very reactive elements in nature.
- \_\_\_\_\_ 30. Elements have properties of both metals and non-metals.

# Chapter 12 Study Guide For Content Mastery

## Stoichimetry

**SB Merriam**



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