

Skills Worksheet

Concept Review

Section: Compounds and Molecules

1. Explain what holds compounds together.

Compounds are held together by chemical bonds.

1. **Electrostatic** forces (attraction) between a full and empty orbital and a pair filling orbital of a compound.

Electrostatic forces are the forces that hold atoms together in a compound. They are the forces that hold atoms together in a compound.

2. **Electrostatic** forces (attraction) between a full and empty orbital and a pair filling orbital of a compound.

Electrostatic forces are the forces that hold atoms together in a compound. They are the forces that hold atoms together in a compound.

3. **Electrostatic** forces (attraction) between a full and empty orbital and a pair filling orbital of a compound.

Electrostatic forces are the forces that hold atoms together in a compound. They are the forces that hold atoms together in a compound.

4. **Electrostatic** forces (attraction) between a full and empty orbital and a pair filling orbital of a compound.

Electrostatic forces are the forces that hold atoms together in a compound. They are the forces that hold atoms together in a compound.

5. **Electrostatic** forces (attraction) between a full and empty orbital and a pair filling orbital of a compound.

Electrostatic forces are the forces that hold atoms together in a compound. They are the forces that hold atoms together in a compound.

Concept Review Section Compounds Of Carbon Answer

Ian Pickup



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