

Chapter 2 The Nature of Matter • Section 1 Summary

Describing Matter

Key Concepts

- What kinds of properties are used to describe matter?
- What are elements, and how do they relate to compounds?
- What are the properties of a mixture?

Matter is anything that has mass and takes up space. **Chemistry** is the study of the properties of matter and how matter changes. In chemistry, a **substance** is a single kind of matter that is pure.

Every form of matter has two kinds of properties—**physical properties** and **chemical properties**. A **physical property** is observed without changing a substance into another substance. Examples of physical properties are hardness, texture, color, and ability to dissolve in water. A **chemical property** is the ability of a substance to change into different substances. Some chemical properties are burning and rusting.

All matter is made up of elements. An **element** is a pure substance that cannot be broken down into any other substance. **Elements** are the simplest substances. Each element is identified by its specific physical and chemical properties. An **atom** is the basic particle that makes up an element. Atoms of most elements can combine with other atoms. A **chemical bond** is the force that holds two atoms together. Atoms often combine to form **molecules**, which are larger particles made of two or more atoms held together by chemical bonds.

When elements are chemically combined, they form **compounds** having properties that are different from those of the uncombined elements. A **compound** is a pure substance made of two or more elements chemically combined in a set ratio. A compound may be represented by a **chemical formula**. A chemical formula shows the elements in the compound and the ratio of atoms. For example, the chemical formula for carbon dioxide is CO_2 . In carbon dioxide, there are always two oxygen atoms to every one carbon atom.

Elements and compounds are pure substances, but most of the materials you see every day are not. Instead, they are mixtures. A **mixture** is made of two or more substances that are together in the same place, but are not chemically combined. Mixtures differ from compounds in two ways. Each substance in a mixture keeps its individual properties. Also, the parts of a mixture are not combined in a set ratio.

A mixture can be heterogeneous or homogeneous. In a **heterogeneous mixture**, you can see the different parts. The substances in a **homogeneous mixture** are so evenly mixed that you cannot see the different parts. A **solution** is an example of a homogeneous mixture. Air is a solution of nitrogen gas, oxygen gas, plus small amounts of other gases. Unlike compounds, mixtures are easily separated into their components. For example, iron filings can be easily removed from salt with a magnet.

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Holt Chemistry Holt Rinehart & Winston, 2003-01-24 *Matter* Prentice-Hall Staff, 1994 *Concepts in Biology'* 2007 Ed. 2007 Edition , *Matter, Building Block of the Universe* , 1993 *Ebook: Chemistry: The Molecular Nature of Matter and Change* Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change **States of Matter, States of Mind** Allan F.M. Barton, 2021-08-29 States of Matter States of Mind is an easy to read introduction to the way the physical world is put together and stays together The book presents the fundamental ideas and particles of the makeup of the universe to enable understanding of matter and why it behaves in the way it does Written in an engaging manner the book explains some of the intricate details and grand schemes of life and the universe by making analogies with common everyday examples For example the recipe for a cake tells us nothing of how good the cake tastes but is a model of the food and a scientific model is no closer to the reality of the materials than a recipe is to the mouth watering flavor of the cake Illustrated with helpful cartoons this book provides a vast knowledge of atoms and atmospheres The first several chapters introduce terms and fundamental ideas while later chapters deal successively with particles and systems from the electron to the universe as a system Each new idea introduced builds upon the last A user friendly bibliography provides references for further reading **Chemistry: Core Concepts, 3rd Edition** Allan Blackman, Daniel Southam, Gwendolyn Lawrie, Natalie Williamson, Christopher Thompson, 2024-01-09 The third edition of Chemistry Core Concepts Blackman et al has been developed by a group of leading chemistry educators for students entering university with little or no background in chemistry Available as a full colour printed textbook with an interactive eBook code this title enables every student to master concepts and succeed in assessment Lecturers are supported with an extensive and easy to use teaching and learning package *Matter and Energy* , 2005 *Topological Disorder in Condensed Matter* F. Yonezawa, T. Ninomiya, 2012-12-06 This volume contains papers presented at the Fifth Taniguchi Symposium on the Theory of Condensed Matter which was held between 2 5 November 1982 at Shimoda Japan The topic of the Symposium was Topological Disorder in Condensed Matter The objective of the Taniguchi Symposium is to encourage activity in those fields of research not in the limelight at the moment but regarded as very promising such as our theme Topological disorder refers to the dis order in the positions and connectivities of atoms in amorphous solids and liquids The development of the physiCS of topologically disorderd systems though extremely important fundamentally and for application purposes falls far behind compared to that of other kinds of disorderd systems because the structure characterization of topologically disordered systems is still at a rather primitive stage The structure characterization is the key to com prehensive understanding of physical properties of any material Recently several new attempts at structural analyses have been reported Encouraged by this fact our motivation in organizing the symposium was to investigate the possibilities of theoretical approaches to open a breakthrough in the present research situation on this subject A rough sketch of the problem is made in the Introduction to give the readers a

general outline of the subject Part I is devoted to several attempts to synthesize and characterize topological disorder more or less by analytical means New Science Discovery for Lower Secondary Rex M. Heyworth, 2008

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