

Acid/Base indicator: _____ that changes _____ in the presence of an _____ or a _____.

Solution	Litmus paper (R→B, B→R, or NC)	Phenolphthalein (color or NC)	Bromothymol Blue (color or NC)	Cabbage Juice (color or NC)
HNO ₃				
NaOH				
KOH				
H ₂ SO ₄				

Conclusion Questions:

- Litmus turns _____ in an acid and _____ in a base.
- Phenolphthalein (phth) turns _____ in an acid and _____ in a base.
- Bromothymol blue turns _____ in an acid and _____ in a base.
- Cabbage juice turns _____ in an acid and _____ in a base.

Strong Acids: dissociate _____ in _____ (ex: _____)
 Weak Acids: dissociate _____ in _____
 (ex: _____ or _____)

Strong Bases: dissociate _____ in _____ (ex: _____)
 Weak Bases: dissociate _____ in _____
 (ex: _____ or _____)

pH = _____

0 _____ 7 _____ 14

Determine the pH of a solution of HCl that has a molarity of $1 \times 10^{-4} \text{ M}$.

Chemistry Note Taking Guide Episode 1102 Key

M Carnoy



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