

Chapter 21

Organic Chemistry: Some Basic Principles and Techniques

1. The IUPAC name of neopentane is [AIEEE-2009]

- (1) 2, 2-dimethylpropane
- (2) 2-methylpropane
- (3) 2, 2-dimethylbutane
- (4) 2-methylbutane

2. Arrange the carbanions, $(\text{CH}_3)_3\text{C}^-$, CCl_3^- , $(\text{CH}_3)_2\text{CH}^-$, $\text{C}_6\text{H}_5\text{CH}_2^-$, in order of their decreasing stability [AIEEE-2009]

- (1) $(\text{CH}_3)_2\text{CH}^- > \text{CCl}_3^- > \text{C}_6\text{H}_5\text{CH}_2^- > (\text{CH}_3)_3\text{C}^-$
- (2) $\text{CCl}_3^- > \text{C}_6\text{H}_5\text{CH}_2^- > (\text{CH}_3)_2\text{CH}^- > (\text{CH}_3)_3\text{C}^-$
- (3) $(\text{CH}_3)_3\text{C}^- > (\text{CH}_3)_2\text{CH}^- > \text{C}_6\text{H}_5\text{CH}_2^- > \text{CCl}_3^-$
- (4) $\text{C}_6\text{H}_5\text{CH}_2^- > \text{CCl}_3^- > (\text{CH}_3)_3\text{C}^- > (\text{CH}_3)_2\text{CH}^-$

3. The alkene that exhibits geometrical isomerism is [AIEEE-2009]

- (1) 2 - methyl propene
- (2) 2 - butene
- (3) 2 - methyl - 2 - butene
- (4) Propene

4. The correct order of increasing basicity of the given conjugate bases ($\text{R} = \text{CH}_3$) is [AIEEE-2010]

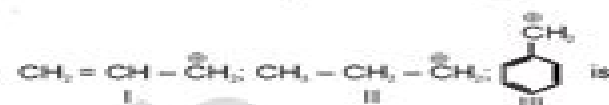
- (1) $\text{RCOO}^- < \text{HC}=\text{C}^- < \text{NH}_2^- < \text{R}^-$
- (2) $\text{RCOO}^- < \text{HC}=\text{C}^- < \text{R}^- < \text{NH}_2^-$
- (3) $\text{R}^- < \text{HC}=\text{C}^- < \text{RCOO}^- < \text{NH}_2^-$
- (4) $\text{RCOO}^- < \text{NH}_2^- < \text{HC}=\text{C}^- < \text{R}^-$

5. 29.5 mg of an organic compound containing nitrogen was digested according to Kjeldahl's method and the evolved ammonia was absorbed in 20 mL of 0.1 M HCl solution. The excess of the acid required 15 mL of 0.1 M NaOH solution for

complete neutralisation. The percentage of nitrogen in the compound is [AIEEE-2010]

- (1) 29.5
- (2) 59.0
- (3) 47.4
- (4) 23.7

6. The order of stability of the following carbocations



[JEE (Main)-2013]

- (1) III > II > I
- (2) II > III > I
- (3) I > II > III
- (4) III > I > II

7. A gaseous hydrocarbon gives upon combustion 0.72 g of water and 3.08 g of CO_2 . The empirical formula of the hydrocarbon is [JEE (Main)-2013]

- (1) C_2H_6
- (2) C_2H_4
- (3) C_6H_6
- (4) C_7H_8

8. For the estimation of nitrogen, 1.4 g of an organic compound was digested by Kjeldahl method and the evolved ammonia was absorbed in 60 mL of

$\frac{\text{M}}{10}$ sulphuric acid. The unreacted acid required 20

mL of $\frac{\text{M}}{10}$ sodium hydroxide for complete neutralization. The percentage of nitrogen in the compound is [JEE (Main)-2014]

- (1) 6%
- (2) 10%
- (3) 3%
- (4) 5%

9. In Carius method of estimation of halogens, 250 mg of an organic compound gave 141 mg of AgBr. The percentage of bromine in the compound is (At. mass Ag = 108; Br = 80) [JEE (Main)-2015]

- (1) 24
- (2) 36
- (3) 48
- (4) 60

Chemistry Ch 21 Study Guide Answers

Neil E. Schore



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Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein, 2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems **Study Guide and Solutions Manual** Neil E. Schore, 2002-08-02 **Organic Chemistry Study Guide and Solutions Manual** Marc Loudon, Jim Parise, Lisa Bonner, 2024-08-09 The Study Guide and Solutions Manual offers several study tools for students Study Guide Links are brief supplements to text material and provide hints for study shortcuts approaches to problem solving or additional explanations of topics that typically cause difficulty Further Explorations are short discussions that provide additional depth on content in the text Reaction Reviews present the reactions in the chapter in detailed outline form This summary consolidates the reactions of the chapter in one place and presents not only the reactions themselves but also the essential features of their mechanisms and stereochemistry for cases in which these issues are discussed in the text Solutions are provided for all of the over 1600 problems in the text Many include worked out examples in which not only the answer is given but also the detailed approach to the solution is discussed *Organic Chemistry, 5e Student Study Guide and Solutions Manual* David R. Klein, Laurie S. Starkey, 2025-03-18 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Study Guide and Solutions Manual 5th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry *Organic Chemistry Study Guide with Solutions Manual* Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer **Study Guide to Accompany Calculus for the Management, Life, and Social Sciences** Clyde Metz, 1984-01-01 Study Guide to Accompany Calculus for the Management Life and Social Sciences "O" Level Study Guide - Chemistry Quite Easily Done , *Organic Chemistry, 4e Student Solution Manual and Study Guide* David R. Klein, 2021-01-07 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Solution Manual and Study Guide 4th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Chemistry, Student Study

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