

Name _____ Date _____

Chapter 3 – Scientific Measurement Study Guide

- Qualitative vs. Quantitative measurements
- Scientific notation → decimal number
 - Ex.) 3.45×10^4
 - Ex.) 6.02×10^9
- Decimal number → scientific notation
 - Ex.) 450,000,000
 - Ex.) .0000000045
- Multiply and divide numbers in scientific notation
 - What are the rules? What do you do with the coefficient? What do you do with the exponent?
 - Ex.) $(5 \times 10^4) \times (1.1 \times 10^6)$
 - Ex.) $(8 \times 10^9) \div (4 \times 10^3)$
- Add and subtract numbers in scientific notation
 - What are the rules? What do you do with the coefficient? What do you do with the exponent?
 - Ex.) $(4 \times 10^7) + (3 \times 10^7)$
 - Ex.) $(2 \times 10^9) - (4 \times 10^9)$
- % error (know how to use the formula)
 - Ex.) If the observed value for a density is 0.80 g/ml and the accepted value is 0.70 g/mL, what is the percent error?

Chemistry Scientific Measurement Study Guide Answers

**Norman E. Griswold, Henry Fuller
Holtzclaw**



Chemistry Scientific Measurement Study Guide Answers:

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Malone,2011-12-07 This is the Study Guide and Solutions Manual to accompany Malone s Basic Concepts of Chemistry

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