

what contents of dirt is responsible for creating a healthy environment:

Introduction

The minerals in dirt are essential for creating a health environment for plants and animals. Minerals in dirt are like the food for use humans. The right dirt for example protein fats vitamins eg for a human can create a strong and healthy person. It's the same for plants. Plants require sulfate as does humans need proteins for building blocks for repairing and building. The sulfate supports creation of plant proteins. So if there is no sulfate the will be less plants less plant means fewer proteins nutrient more pollutants more global warming more water more water more drowning. This may lead to a green house run off affect possible wiping the planet clean off earth. It can be said the more minerals in the dirt the better for all inhabitants of earth. Anyhow soil can be used to find places to build buildings, dams and farming and much more. $\text{Ba}^{2+} + \text{SO}_4^{2-} \rightarrow \text{BaSO}_4$

Aim

To find the percentage of sulfate within the samples of dirt with the equation $\text{Ba} + \text{SO}_4 \rightarrow \text{BaSO}_4$ to find the values of the sulfate with

Hypothesis

The dirt with the darker and the highest moisture and is found in a higher biodiversity environment before the sample was collected will have the high sulfate levels. So sample 1000

Sample 100 from a moist forest like conditions on Blandong

plenty of plants around

Sample 101 from a dry forest a part like condition in know district

tree roots near it

location and time

10

10:00am to 11:00

Risk assessment

when handling

hydrochloric acid make sure you pour slowly if hydrochloric acid has made contact with skin or clothing wash location of contact

barium chloride is harmful to swallow so wash hands

the Bunsen burner is hot don't touch with bare hands

the large beaker is hot after the boil don't use hands to move

wear safety glasses and laboratory coats

variables

independent variables

barium

chloride

amount of dirt sample

hydrochloric acid

dependent variables

sulfate

deionised water

RESULTS

number of grams of original sample	starting weight	end weight	difference	mass of precipitate that is sulfate	mass of sulfur	mass of dirt sample	percentage of sample that is sulfate
100	376.875	376.79	0.085	0.04802	0.00844	1.0001	0.844
101	0.639	0.638	0.009	0.00897	0.00023	1	0.023

- Materials and equipment
- Dirt or fertilizer sample
- 1.0g
- 2ml of hydrochloric acid
- 2ml
- 10ml 0.5 barium chloride solution
- deionised water
- beakers 2000ml, 5000ml
- funnel 2l
- large funnel
- 10ml measuring funnel
- Burette and burette stand
- Filter paper
- Small filter paper
- Filter funnel
- Side arm conical flask
- Safety equipment science coat and safety glasses
- Marker and pencil
- Shoring rod
- Large white tile
- Bunsen burner
- heat probe mat
- glass mat
- tripod stand
- clamp
- weights

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**John Krenos, Joseph Potenza, Carl
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