

Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 4

Q1. Calculate the correct number of significant figures: $4.26 - (15.635/5.0)$

- a.) 1.16
- b.) 1.12
- c.) 1.2
- d.) 1.133

Q2. What amount of dioxygen (in gram) contains 1.8×10^{22} molecules?

- a.) 0.960
- b.) 96.0
- c.) 0.0960
- d.) 9.60

Q3. If we consider % in lace of $1/12$ mass of carbon atom is taken to be the relative atomic mass unit, the mass of one mole of a substance will–

- a.) Decrease twice
- b.) Increase two folds
- c.) Remains unchanged
- d.) Be a fraction of the molecular mass of the substance

Q4. The molarity of a solution obtained by mixing 800 mL of 0.5 M HCl with 200 mL of 1 M HCl will be ____.

- a.) 0.8 M
- b.) 0.6 M
- c.) 0.4 M
- d.) 0.2 M

Q5. The number of molecules present in 1 ml of water is ____.

- a.) 1
- b.) 1000
- c.) 2.69×10^{23}
- d.) 6.02×10^{23}

Q6. Define the atomic mass of an element.

Chemistry 11 Exam Questions And Answers

Lei Shi



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Table of Contents Chemistry 11 Exam Questions And Answers

1. Understanding the eBook Chemistry 11 Exam Questions And Answers
 - The Rise of Digital Reading Chemistry 11 Exam Questions And Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 11 Exam Questions And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 11 Exam Questions And Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 11 Exam Questions And Answers
 - Personalized Recommendations
 - Chemistry 11 Exam Questions And Answers User Reviews and Ratings
 - Chemistry 11 Exam Questions And Answers and Bestseller Lists
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 - Highlighting and Note-Taking Chemistry 11 Exam Questions And Answers
 - Interactive Elements Chemistry 11 Exam Questions And Answers
 8. Staying Engaged with Chemistry 11 Exam Questions And Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry 11 Exam Questions And Answers
 9. Balancing eBooks and Physical Books Chemistry 11 Exam Questions And Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 11 Exam Questions And Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Chemistry 11 Exam Questions And Answers
 - Setting Reading Goals Chemistry 11 Exam Questions And Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Chemistry 11 Exam Questions And Answers
 - Fact-Checking eBook Content of Chemistry 11 Exam Questions And Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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 - Integration of Multimedia Elements
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