

Chemistry 530 Midterm Review

Exam on Monday Dec 16 at 8:30

Section I = 75 pts

- Consists of 75 multiple choice questions worth 1 pt. each
- Questions cover all material covered during the second semester
- Questions will be similar to multiple choice problems in the book and review guides on-line
- Practicing the review guides is the best approach; answers will be posted on-line

Section II = 95 pts

- Consists of 19 short answer problems totaling 95 points
- Questions cover the whole semester
- Examples of types of questions
 - Metric Conversions
 - Volume Calculations
 - Percent Composition
 - Atomic Mass
 - Electron configuration
 - Nomenclature
 - Balancing Equations
 - Predicting the products of a reactions
 - Identifying types of reactions
 - Mole Conversions
 - Molar Mass of a compound
 - Molar Mass of a gas at STP
 - Density of a gas at STP
 - Empirical & Molecular Formula

Study approach

- First, read through your notes. Once thoroughly and once more quickly.
- Go through the review packet and make sure you are comfortable with all of the terms and how to do problems
- Do the review packets for help with multiple choice
- Do questions on old tests for practicing written problems
- Make sure you have practiced AT LEAST one of the types of example problems

Chapter 1 – Introduction to Chemistry

- Chemistry: the branch of science that studies the composition and properties of matter.
- Scientific Method:
 - Observation
 - Hypothesis
 - Controlled Experiment
- Theory versus Natural Law

Prerequisite Science Skills

- Rules for Significant Figures:
 - 1) All non-zero digits are significant.
 - 2) Zeros are significant only when they are *between* two non-zero digits or if they occur *after* a decimal point and *after* a non-zero digit.
 - 3) In *multiplication or division*, you cannot end up with more significant figures than you started with.
 - 4) In *addition or subtraction*, you cannot have more digits to the *right* of the decimal than you started with.
- Exponential numbers
- Scientific Notation = $D.DD \times 10^n$ where $1.00 \leq D.DD \leq 9.99$

Chemistry Midterm Study Guide

Puja Mehta



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