

Exercises

2. Following the rule you established in the last key question, write correct chemical formula for each of the following compounds

Compound	Formula
(a) Lithium chloride	$\text{Li}^{+1} / \text{Cl}^{-1} \rightarrow \text{LiCl}$
(b) Magnesium iodide	$\text{Mg}^{+2} / \text{I}^{-1} \rightarrow \text{MgI}_2$
(c) Strontium selenide	$\text{Sr}^{+2} / \text{Se}^{-2} \rightarrow \text{SrSe}$
(d) Rubidium fluoride	$\text{Rb}^{+1} / \text{F}^{-1} \rightarrow \text{RbF}$
(e) Lithium oxide	$\text{Li}^{+1} / \text{O}^{-2} \rightarrow \text{Li}_2\text{O}$
(f) Sodium sulfide	$\text{Na}^{+1} / \text{S}^{-2} \rightarrow \text{Na}_2\text{S}$
(g) Potassium chloride	$\text{K}^{+1} / \text{Cl}^{-1} \rightarrow \text{KCl}$
(h) Calcium phosphide	$\text{Ca}^{+2} / \text{P}^{-3} \rightarrow \text{Ca}_3\text{P}_2$
(i) Barium oxide	$\text{Ba}^{+2} / \text{O}^{-2} \rightarrow \text{BaO}$
(j) Aluminum sulfide	$\text{Al}^{+3} / \text{S}^{-2} \rightarrow \text{Al}_2\text{S}_3$

3. Use your answers to the Key Questions and the Exercise Questions to draw a conclusion about the ratio of ions in two compounds if the elements in the compounds are from the same groups (example: aluminum oxide and aluminum sulfide; lithium chloride and potassium chloride).

Chemistry A Bonding Packet Answers

**Jit Chakraborty & Asimesh Dutta
Gupta and Ravikanth Kamlekar**



Chemistry A Bonding Packet Answers:

Engineering Chemistry-I: Concepts and Applications Jit Chakraborty & Asimesh Dutta Gupta and Ravikanth Kamlekar, Engineering Chemistry I Concepts and Applications is a textbook that offers an exclusive coverage of the topics and proper explanation of concepts as per the present day and future needs of the students The book provides the theoretical Chapters 1-7 as well as practical Chapter 8 aspects of the paper Chemistry I BSC102 as per the latest AICTE curriculum It will be useful to not only the first year engineering and technology students of all streams but also the professors for guiding their students

Essays in Contemporary Chemistry Gerhard Quinkert, 2001 In this volume internationally renowned chemists recount their roles in the progress of chemistry research toward elucidation of biological processes Beginning with a historical perspective on the development of X-ray crystallography the reader is regaled with first hand accounts of research milestones Included are descriptions of the cutting edge nuclear magnetic resonance and electron paramagnetic resonance spectroscopic techniques the dynamics of ultra fast reactions and the central problem of molecular oxygen activation in biological processes The roles of chiral auxiliaries in organic synthesis and of photochemistry in natural product synthesis are told and innovations in combinatorial synthesis approaches are described Contemporary issues in molecular recognition and modulation of molecular function are addressed concluding with a missive regarding how the frontiers of medical ethics may be breached by molecular manipulations The contributors who number among the finest scientists in the world including two Nobel Prize winners are Peter B Dervan Jack D Dunitz Christian Griesinger Jean Marie Lehn Thomas F Prisner Gerhard Quinkert Peter G Schultz Helmut Schwarz Dieter Seebach and Ahmed Zewail Additionally there is a prologue by Albert Eschenmoser for whom this collection was conceived and an epilogue that contains facsimiles of notes from his landmark lecture Synthesis of Co Enzyme B12 A Vehicle for Teaching Organic Synthesis This book is definitely a must for all who want to read or to read again where we stand in our chemical comprehension of the fascinating relationship between chemical structure and biological processes how we got here and what the future might hold

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