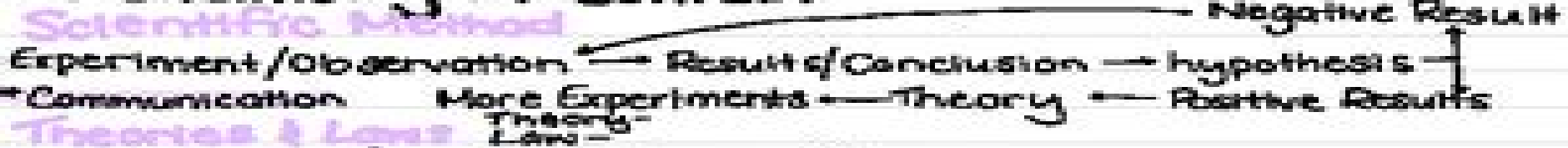


Finals Grinding

Chapter 1

1.1 Chemistry in Context



1.2 Phases & Matter Classification

States of Matter

- Solid - fixed shape and volume
- Liquid - shape of container, fixed volume
- gas - expands to fill container
- plasma - gaseous state w/ charged particles w/ unique properties

Mass & Weight

- Mass - amt of matter
- Weight - force of gravity on object

Conservation

- Physical reactions: Amt of matter doesn't change due to chemical or physical reactions
- Pure substance - constant composition
- Mixture - two or more substances

Defined Terms

- Elements - same type of atom, smallest version
- Compounds - can be decomposed to other compounds/elements
- Homogeneous mixture - uniform composition
- Heterogeneous mixture - physically distinct parts

Miscible - forms a homogenous mixture when mixed

1.3 Physical & Chemical Properties

Ex/Intensive Properties

- extensive - depend on amt
- intensive - independent of amt

Periodic Table

- Atomic number = protons
- Periods - across
- Groups - down

Chalcogens - group 16 (O, S, Se, Te, Po, Lv) Metals ~ metalloids ~ Non ~ Hala ~ Noble gases

alkali ~ alkaline earth ~ transition ~ metals ~ metalloids ~ Non ~ Hala ~ Noble gases

Strong Acids

chloric	HClO_3	hydrobromic	HBr	nitric	HNO_3	hydrochloric	HCl	sulfuric	H_2SO_4
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Strong Bases

barium	Ba(OH)_2	calcium	Ca(OH)_2	strontium	Sr(OH)_2	lithium	LiOH	potassium	KOH	sodium	NaOH
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1.4 Measurements

magnitude - size of number **unit** - standard of comparison

indication of uncertainty - + or - uncertain digit

Pico 10^{-12} nano 10^{-9} micro 10^{-6} milli 10^{-3} centi 10^{-2} deci 10^{-1}

Kilo 10^3 mega 10^6

Density mass/volume (intrinsic property)

1.5 Uncertainty, Accuracy, Precision

measurement device + user introduce uncertainty

measure from bottom of meniscus

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