

Question 8:

An atom X has 2, 8, 7 electrons in its shell. It combines with Y having 1 electron in its outermost shell.

(a) What type of bond will be formed between X and Y?

(b) Write the formula of the compound formed.

Solution 8:

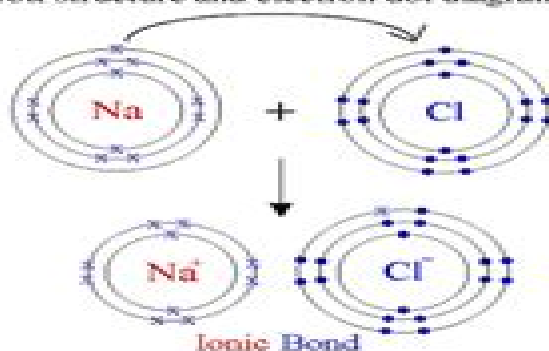
- (a) X has 7 electrons in its outermost shell and Y has only one electron in its outermost shell so Y loses its one electron and X gains that electron to form an ionic bond.
- (b) The formula of the compound would be XY.

Question 9:

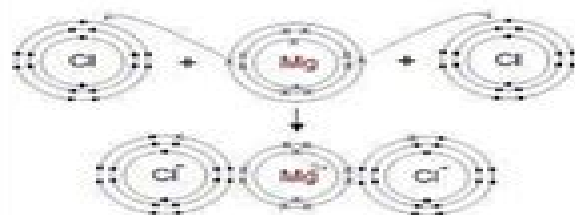
Draw orbit structure and electron dot diagram of NaCl, MgCl₂ and CaO.

Solution 9:

Orbit structure and electron dot diagram of NaCl:



Orbit structure and electron dot diagram of MgCl₂:



Orbit structure and electron dot diagram of CaO:

Concise Chemistry Chemical Bonding Chapter 2 Answers

Yicheng Fang



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Concise Polymeric Materials Encyclopedia Joseph C. Salamone, 1998-08-28 Concise Polymeric Materials Encyclopedia culls the most used widely applicable articles from the Polymeric Materials Encyclopedia more than 1 100 and presents them to you in a condensed well ordered format Featuring contributions from more than 1 800 scientists from all over the world the book discusses a vast array of subjects related to the synthesis properties and applications of polymeric materials development of modern catalysts in preparing new or modified polymers modification of existing polymers by chemical and physical processes biologically oriented polymers This comprehensive easy to use resource on modern polymeric materials serves as an invaluable addition to reference collections in the polymer field

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Science

Books, 1974 **Facets of Coordination Chemistry** B. V. Agarwala, K. N. Munshi, 1993 A concise account of coordination chemistry since its inception is given here together with some of the newer significant facets This book covers a broad spectrum of various topics on Environment Cyclic Voltammetry Chromatography Metal Complexes of biological interest Alkoxides NMR spectroscopy and others These are useful to the scientific community engaged in the field of Inorganic Chemistry and Analytical Chemistry **Chemistry for Degree Students B.Sc. (Honours) Semester I** Madan R.L., 2022 This textbook has been designed to meet the needs of B Sc Honours First Semester students of Chemistry as per the UGC Choice Based Credit System CBCS Maintaining the traditional approach to the subject this textbook lucidly explains the basics of Inorganic and Physical Chemistry Important topics such as atomic structure periodicity of elements chemical bonding and oxidation reduction reactions gaseous state liquid state solid state and ionic equilibrium are aptly discussed to give an overview of inorganic and physical chemistry Laboratory work has also been included to help students achieve solid conceptual understanding and learn experimental procedures *Chemistry of Natural Products* V.K. Ahluwalia, Lalita S. Kumar, Sanjiv Kumar, 2022-11-30 This book is designed to serve as a textbook for core as well as elective courses offered to undergraduate and advanced undergraduate students enrolled in chemistry This textbook comprehensively deals various topics of organic chemistry such as amino acids peptides proteins and enzymes The text is divided into four chapters a chapter each dedicated to amino acids peptides proteins and enzymes respectively The important reactions have been explained with the help of the mechanisms involved It gives a detailed account of the solution phase and solid phase synthesis of peptides as well as discussing the structure and function of some biologically important peptides It also covers the classification nomenclature and mode of action of enzymes and a detailed account of the structure and function of different co enzymes The book also includes pedagogical features like end of chapter exercises to aid in self learning Given the scope this textbook will be useful for graduate and advanced graduate students pursuing the course of chemistry especially organic chemistry **Forensic Chemistry** Suzanne Bell, 2022-04-27 Forensic Chemistry Third Edition the new edition of this ground breaking book continues to serve as the leading forensic chemistry text on the market Fully updated this edition describes the latest advances in current forensic chemistry analysis and practice New and expanded coverage includes rapid advances in forensic mass spectrometry NMR and novel psychoactive substances NPSs Topics related to seized drug analysis toxicology combustion and fire investigation explosives and firearms discharge residue are described and illustrated with case studies The role of statistics quality assurance quality control uncertainty and metrology are integrated into all topics More pharmacological and toxicokinetic calculations are presented and discussed Hundreds of color figures nearly 450 total along with graphs illustrations worked example problems and case descriptions are used to show how analytical chemistry is applied to forensic practice Coverage offer students insight into the legal context in which forensic chemistry is conducted and introduces them to the sample types and sample matrices frequently encountered in

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