Acid-Base Chemistry

acids

- pH less than 7
- Turn litmus paper red
- Taste sour
- Feel irritating/corrosive
- Include fruit juices, soda, and coffee

bases

- pH greater than 7
- · Turn litmus paper blue
- Taste bitter or soapy
- · Feel slippery
- Include baking soda, ammonia, and soap



neutralization reaction example: HCl + NaOH → NaCl + H₂O

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Nick Kanellopoulos

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Inorganic Chemistry For Dummies Michael Matson, Alvin W. Orbaek, 2013-06-24 The easy way to get a grip on inorganic chemistry Inorganic chemistry can be an intimidating subject but it doesn t have to be Whether you re currently enrolled in an inorganic chemistry class or you have a background in chemistry and want to expand your knowledge Inorganic Chemistry For Dummies is the approachable hands on guide you can trust for fast easy learning Inorganic Chemistry For Dummies features a thorough introduction to the study of the synthesis and behavior of inorganic and organometallic compounds In plain English it explains the principles of inorganic chemistry and includes worked out problems to enhance your understanding of the key theories and concepts of the field Presents information in an effective and straightforward manner Covers topics you ll encounter in a typical inorganic chemistry course Provides plain English explanations of complicated concepts If you re pursuing a career as a nurse doctor or engineer or a lifelong learner looking to make sense of this fascinating subject Inorganic Chemistry For Dummies is the quick and painless way to master inorganic chemistry

The Pearson Complete Guide for the AIEEE 2012 Dinesh Khattar, Ravi Raj Dudeja, K.K. Arora,

Comprehensive Inorganic Chemistry II, 2013-07-23 Comprehensive Inorganic Chemistry II Nine Volume Set reviews and examines topics of relevance to today s inorganic chemists Covering more interdisciplinary and high impact areas Comprehensive Inorganic Chemistry II includes biological inorganic chemistry solid state chemistry materials chemistry and nanoscience The work is designed to follow on with a different viewpoint and format from our 1973 work Comprehensive Inorganic Chemistry edited by Bailar Emel us Nyholm and Trotman Dickenson which has received over 2 000 citations The new work will also complement other recent Elsevier works in this area Comprehensive Coordination Chemistry and Comprehensive Organometallic Chemistry to form a trio of works covering the whole of modern inorganic chemistry Chapters are designed to provide a valuable long standing scientific resource for both advanced students new to an area and researchers who need further background or answers to a particular problem on the elements their compounds or applications Chapters are written by teams of leading experts under the guidance of the Volume Editors and the Editors in Chief The articles are written at a level that allows undergraduate students to understand the material while providing active researchers with a ready reference resource for information in the field The chapters will not provide basic data on the elements which is available from many sources and the original work but instead concentrate on applications of the elements and their compounds Provides a comprehensive review which serves to put many advances in perspective and allows the reader to make connections to related fields such as biological inorganic chemistry materials chemistry solid state chemistry and nanoscience Inorganic chemistry is rapidly developing which brings about the need for a reference resource such as this that summarise recent developments and simultaneously provide background information Forms the new definitive source for researchers interested in elements and their applications completely replacing the highly cited first edition which

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