

Enzyme and Temperature

Background

For the experiment an enzyme (catalase) breaking down hydrogen peroxide when heated will be used and tested.

Introduction

Enzymes are proteins that increase the rate of a chemical reaction (catalysis). They are made of amino acids and are specific to a reaction. Enzymes are affected by temperature, pH, and substrate concentration. In this experiment, the effect of temperature on the activity of the enzyme catalase will be tested.

Key Concepts

Enzymes are biological catalysts that speed up chemical reactions.

Variables

Independent	Dependent	Controlled
Temperature of the reaction	Rate of reaction (measured by volume of oxygen gas produced)	Amount of enzyme, amount of substrate, pH of the reaction

Procedure

Step 1	Step 2	Step 3
		
Preparation of enzyme solution	Addition of substrate	Measurement of reaction rate

1. Preparation of enzyme solution: A solution of catalase enzyme is prepared by adding 10 mL of distilled water to 10 mL of 1% catalase solution in a test tube. The mixture is then shaken well.

2. Addition of substrate: 5 mL of 3% hydrogen peroxide solution is added to the enzyme solution in the test tube. The reaction is observed and the volume of oxygen gas produced is measured.

3. Measurement of reaction rate: The volume of oxygen gas produced is measured by the displacement of water in an inverted graduated cylinder. The reaction is observed and the volume of oxygen gas produced is measured.

Results

Temp (°C)	Volume of O ₂ (mL)	Rate of Reaction (mL/min)
10	1.0	0.1
20	2.0	0.2
30	3.0	0.3
40	4.0	0.4
50	5.0	0.5
60	6.0	0.6

Discussion

Temp (°C)	Volume of O ₂ (mL)	Rate of Reaction (mL/min)
10	1.0	0.1
20	2.0	0.2
30	3.0	0.3
40	4.0	0.4
50	5.0	0.5
60	6.0	0.6

As the temperature of the reaction increases, the rate of reaction also increases. This is because the kinetic energy of the molecules increases, leading to more frequent collisions between the enzyme and the substrate.



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