

COMMON NAME AND FORMULA OF IMPORTANT CHEMICAL COMPOUNDS

Common Name	Chemical Compound	Chemical Formula
Baking Powder	Sodium Bicarbonate	NaHCO_3
Bleaching Powder	Calcium Oxychloride	CaOCl_2
Blue Vitriol	Copper Sulphate	$\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$
Caustic Potash	Potassium Hydroxide	KOH
Caustic Soda	Sodium Hydroxide	NaOH
Chalk (Marble)	Calcium Carbonate	CaCO_3
Chloroform	Trichloro Methane	CHCl_3
Dry Ice	Solid Carbondioxide	CO_2
Epsom	Magnesium Sulphate	MgSO_4
Green Vitriol	Ferrous Sulphate	FeSO_4
Gypsum	Calcium Sulphate	CaSO_4
Heavy Water	Deuterium Oxide	D_2O
Laughing Gas	Nitrous Oxide	N_2O
Magnesia	Magnesium Oxide	MgO
Marsh Gas	Methane	CH_4
Mohr's Salt	Ammonjum Ferrous Sulphate	$\text{FeSO}_4(\text{NH}_4)_2\text{SO}_4 \cdot 6\text{H}_2\text{O}$
Plaster of Paris	Calcium Sulphate	$\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$
Potash Alum	Potassium Aluminium Sulphate	KAlSO_4
Quick Lime	Calcium Oxide	CaO
Sand	Silicon Oxide	SiO_2
Slaked Lime	Calcium Hydroxide	Ca(OH)_2
Sugar	Sucrose	$\text{C}_{12}\text{H}_{22}\text{O}_{11}$
T.N.T.	Trinitrotoluene	$\text{C}_7\text{H}_5\text{N}_3\text{O}_6$
Vermeljum	Mercuric Sulphide	HgS
Vinegar	Acetic Acid	CH_3COOH
Washing Soda	Sodium Carbonate	Na_2CO_3
White Vitriol	Zinc Sulphate	$\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$

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Arshad Iqbal



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chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students **O**

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web the image below depicts the pin locations as if viewing the connector on the ecu the image below depicts the 3b ecu pinout when viewing the cable assembly the following table defines the functionality of each of the 55 pins on the 3b motronic ecu

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