## Solution Concentration

### Study Guide

## **Key Concepts**

- Concentrations can be measured qualitatively and quantitatively.
- Molarity is the number of moles of solute dissolved per liter of solution.

molarity 
$$(M) = \frac{\text{moles of solute}}{\text{liters of solution}}$$

 Molality is the ratio of the number of moles of solute dissolved in 1 kg of solvent.

molality 
$$(m) = \frac{\text{moles of solute}}{\text{kg of solvent}}$$







# **Chemistry Solution Concentration Study Guide Answer Key**

**Huangqi Zhang** 

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