

G. C. 2-9: LABORATORY REPORT

Name: Kassidy Heinz

YOU MUST SHOW ALL YOUR CALCULATIONS TO RECEIVE CREDIT FOR THEM!

Part I. Addition of 0.100 M HCl to Buffer and Deionized Water

Concentration of buffer: For acetic acid: 0.1 M For acetate: 0.1 M

Volume of HCl	Total Volume of HCl	pH Meter Reading		Expected pH (Calculated)	
		Buffer	Water	Buffer	Water
0	0	4.76	7.00	4.76	
5.00	5.00	4.67	2.04		
5.00	10.00	4.58	1.78		
10.00	20.00	4.39	1.54		
10.00	30.00	4.16	1.43		
20.00	50.00	2.88	1.30		
50.00	100.00	1.48	1.18		

Calculations must be shown in your laboratory notebook.

Part II. Addition of 0.100 M NaOH to Buffer and Deionized water

Volume of NaOH	Total Volume NaOH	pH Meter Reading		Expected pH (Calculated)	
		Buffer	Water	Buffer	Water
0	0	4.76	7		
5.00	5.00	4.84	11.95		
5.00	10.00	4.93	12.22		
10.55	20.00	5.13	12.45		
10.00	30.00	5.36	12.52		
20.00	50.00	8.88	12.69		
50.00	100.00	12.51	12.82		

Calculations must be shown in your laboratory notebook.

Completed Lab 2046 Manual

**Maxie Nave Woodring, Mervin Elijah
Oakes, Henry Emmett Brown**

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