

Compounds and Their Bonds

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A. Ionic Transfer of Electrons

Element	Atomic #	1. Electron Configuration of Atom	2. Electron-Dot Symbol	3. Loss or Gain	4. Electron Configuration of Ion	5. Ionic Charge	6. Symbol of Ion	7. Name of Ion
Sodium	11	$1s^2 2s^2 2p^6 3s^1$	$\cdot \overset{\cdot}{\underset{\cdot}{\text{Na}}} \cdot$	Loss 1 e^-	$1s^2 2s^2 2p^6$	1 ⁺	Na^+	Sodium ion
Nitrogen	7	$1s^2 2s^2 2p^3$	$\cdot \overset{\cdot}{\underset{\cdot}{\text{N}}} \cdot$	Gain 3 e^-	$1s^2 2s^2 2p^6$	3 ⁻	N^{3-}	Nitride ion
Aluminum	13	$1s^2 2s^2 2p^6 3s^2 3p^1$	$\cdot \overset{\cdot}{\underset{\cdot}{\text{Al}}} \cdot$	Loss 3 e^-	$1s^2 2s^2 2p^6$	3 ⁺	Al^{3+}	Aluminum ion
Chlorine	17	$1s^2 2s^2 2p^6 3s^2 3p^5$	$\cdot \overset{\cdot}{\underset{\cdot}{\text{Cl}}} \cdot$	Gain 1 e^-	$1s^2 2s^2 2p^6 3s^2 3p^6$	1 ⁻	Cl^-	Chloride ion
Calcium	20	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2$	$\cdot \overset{\cdot}{\underset{\cdot}{\text{Ca}}} \cdot$	Loss 2 e^-	$1s^2 2s^2 2p^6 3s^2 3p^6$	2 ⁺	Ca^{2+}	Calcium ion
Oxygen	8	$1s^2 2s^2 2p^4$	$\cdot \overset{\cdot}{\underset{\cdot}{\text{O}}} \cdot$	Gain 2 e^-	$1s^2 2s^2 2p^6$	2 ⁻	O^{2-}	Oxide ion

B. Ionic Compounds and Formulas

1. Physical properties

Compound	Appearance	Density	Melting Point
Sodium chloride, NaCl	White, solid, crystalline	2.16 g/cm ³	801°C

2. Formulas of ionic compounds

Name	Positive Ion	Negative Ion	Formula
Lithium iodide	Li^{+1}	I^{-1}	LiI
Aluminum oxide	Al^{3+}	O^{2-}	Al_2O_3
Calcium sulfide	Ca^{2+}	S^{2-}	CaS
Magnesium bromide	Mg^{2+}	Br^{-1}	MgBr_2
Potassium nitride	K^{+1}	N^{3-}	K_3N
Sodium fluoride	Na^{+1}	F^{-1}	NaF

3. Names of ionic compounds

Formula	Positive Ion	Negative Ion	Name
K_2S	K^{+1}	S^{2-}	Potassium sulfide
BaF_2	Ba^{2+}	F^{-1}	Barium fluoride
MgO	Mg^{2+}	O^{2-}	Magnesium oxide
Na_3N	Na^{+1}	N^{3-}	Sodium nitride
AlCl_3	Al^{3+}	Cl^{-1}	Aluminum chloride
Mg_3P_2	Mg^{2+}	P^{3-}	Magnesium phosphide

Compounds Their Formulas Lab 7 Answers

Jo Allan Beran



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In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Compounds Their Formulas Lab 7 Answers**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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