

2002 Edition • Reference Tables for Chemistry Scavenger Hunt

Directions : Your reference tables will be your most valuable tool during the course of Regents Chemistry. They contain an amazing amount of information! The problem is, many students are not well enough acquainted with their tables and are unaware that the answers to many questions are literally at their fingertips! To that end, this activity is a scavenger hunt designed to lead you in the use of every table in the 2002 reference tables for Chemistry. Answer the questions on the left, and write the letter and title of the table used on the right. Good luck and happy hunting!

QUESTION

TABLE /TITLE

1. Complete the chart below with the missing labels.

KEY	At. Mass 12.0111 -4 Symbol C +2 +4 at. # 6 e⁻ conf 2-4	<i>Selected Ox. state</i> Relative atomic masses are based on $^{12}\text{C} = 12.000$ Note: Mass numbers in parentheses are mass numbers of the most stable or common isotope.
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2. At a temperature of 75°C , what is the vapor pressure of ethanol?
85 kPa
3. List the conditions present at STP (standard temperature and pressure)

Table H - Vapor pressure

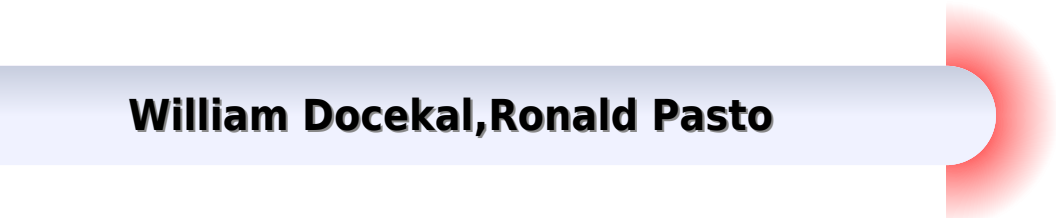
Table A - STP

STP pressure : 101.3 atm
101.3 kPa

STP temp : 273 K
0 $^{\circ}\text{C}$

Chemistry Reference Table Scavenger Hunt Answers

William Docekal, Ronald Pasto



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