

Chemistry, 6e (McMurry/Fay)**Chapter 3 Formulas, Equations, and Moles****3.1 Multiple-Choice Questions**

1) Chemical equations are balanced in order to obey the law of

- A) definite proportions.
- B) mass action.
- C) mass conservation.
- D) multiple proportions.

Answer: C

Topic: Section 3.1 Balancing Chemical Equations

2) Which one of the following statements about balanced equations is **false**?

In a balanced reaction

- A) atoms must be balanced on both sides of the reaction arrow.
- B) mass must be conserved.
- C) molecules must be balanced on both sides of the reaction arrow.
- D) net charge must be balanced on both sides of the reaction arrow.

Answer: C

Topic: Section 3.1 Balancing Chemical Equations

3) Which one of the following statements about balanced equations is true? A reaction is balanced by

- A) changing the charge on an ion.
- B) changing the formula of the molecule.
- C) multiplying by suitable coefficients.
- D) rearranging atoms in a molecule.

Answer: C

Topic: Section 3.1 Balancing Chemical Equations

4) What is the stoichiometric coefficient for oxygen when the following equation is balanced using the lowest whole-number coefficients



- A) 3
- B) 5
- C) 7
- D) 9

Answer: D

Topic: Section 3.1 Balancing Chemical Equations

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