

Matter—Properties and Changes

Section 3.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical
density

mass
properties

physical
substance

Matter is anything with **(1)** _____ and volume. A **(2)** _____ is a form of matter with a uniform and unchanging composition. Substances have specific, unchanging **(3)** _____ that can be observed. Substances have both physical and chemical properties. **(4)** _____ properties can be observed without changing a substance's chemical composition. Color, hardness, and **(5)** _____ are examples. Other properties cannot be observed without changing the composition of a substance. These are called **(6)** _____ properties. An example is the tendency of iron to form rust when exposed to air.

Label each property as either *physical* or *chemical*.

- _____ 7. Chemical formula H_2O
- _____ 8. Forms green carbonate when exposed to moist air
- _____ 9. Remains unchanged when in the presence of nitrogen
- _____ 10. Colorless
- _____ 11. Solid at normal temperatures and pressures
- _____ 12. Ability to combine with another substance
- _____ 13. Melting point
- _____ 14. Liquid at normal temperatures and pressures
- _____ 15. Boiling point is $100^{\circ}C$
- _____ 16. Conducts electricity
- _____ 17. Density is $\frac{1g}{cm^3}$

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