

Acids, Bases, Solutions Study Guide

4 pages + 4-page Answer Key

chemistry

Physical Science

SOLUTIONS, ACIDS, AND BASES STUDY GUIDE

What is a heterogeneous mixture? Provide two examples.

Describe the characteristics of a saturated solution and an unsaturated solution.

Define the term solvent and provide an example.

What is solubility, and how is it measured?

Describe the difference between concentrated and dilute.

What are the properties of acids?

What are the properties of bases?

What is the pH scale?

What is the difference between a strong acid and a weak acid?

What is the difference between a strong base and a weak base?

What is the difference between a strong electrolyte and a weak electrolyte?

What are the properties of bases and how are they measured?

Explain the significance of the pH scale.

How does the concentration of an acid affect its pH?

What happens to the pH of a solution during neutralization?

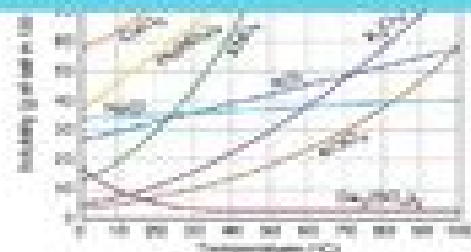
What happens to the pH of a solution when a strong acid and a strong base are mixed?

What happens to the pH of a solution when a strong acid and a weak base are mixed?

What happens to the pH of a solution when a weak acid and a strong base are mixed?

What happens to the pH of a solution when a weak acid and a weak base are mixed?

What happens to the pH of a solution when a strong electrolyte and a weak electrolyte are mixed?



20 Describe how you would use the graph to determine the amount of Na_2SO_4 that can be dissolved in 100 grams of water at 80°C.

22 Which salt has the highest solubility at 80°C?

ANSWER KEY INCLUDED



Chemistry Acids And Bases Study Guide

JR Anderson



Chemistry Acids And Bases Study Guide:

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