Chemical Reaction Example Oxided Practice Problems

Directions: Write balanced chemical equations for each of the following reactions. Be sure to include phases of matter.

- 1. Solid iron (III) oxide reacts with gaseous hydrogen chloride to form iron (III) chloride and water vapor.
- Nitric acid dissolved in water reacts with solid sodium carbonate to form liquid water and carbon dioxide gas and sodium nitrate dissolved in water.

Convert these equations to sentences

- 3. $Fe_{(r)}$ + $O_{2(g)}$ \rightarrow $Fe_2O_{3(r)}$
- 4. $Cu_{(0)}$ + $AgNO_{X(N)}$ \rightarrow $Ag_{(0)}$ + $Cu(NO_1)_{X(N)}$
- NO_{3(d)} → N_{3(d)} + O_{3(d)}

Additional Practice

- Silver nitrate solution and potassium chromate solution, when mixed together, produce solid silver chromate and potassium nitrate solution.
- Phosphoric acid solution reacts with barium chloride solution to produce solid barium phosphate and hydrochloric acid solution.
- 8. Solid magnesium burns in air to produce magnesium oxide, a solid white powder.
- 9. When solid sulfur is exposed to gaseous fluorine, sulfur hexafluoride gas is formed.
- When exposed to manganese (IV) oxide, hydrogen peroxide rapidly breaks down into water and oxygen gas.
- Solid sodium carbonate reacts with hydrochloric acid solution to produce carbon dioxide gas, water and sodium chloride solution.

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Navneet Singh

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colleagues and parents to create a holistic support system for students I collaborate with fellow teachers to share best practices and coordinate curriculum I also maintain open communication with parents through regular updates conferences and discussions about student progress and goals What do you see as the biggest challenge facing education today Answer One of the biggest challenges is ensuring equity and accessibility in education Addressing disparities in resources supporting diverse student needs and fostering inclusive learning environments are critical to ensuring all students have equal opportunities to succeed How do you stay current with educational trends and best practices Answer I actively participate in professional development workshops conferences and online courses to stay updated on educational research trends and innovative teaching strategies I also collaborate with colleagues and engage in reflective practices to continuously improve my teaching What are your strengths as a teacher Answer My strengths include strong communication skills adaptability to diverse student needs creativity in lesson planning and a commitment to fostering a positive classroom environment I am also dedicated to ongoing professional growth and development to enhance my teaching practices Why do you think you would be a good fit for our school district Answer I am drawn to your school district s commitment to mention specific values or initiatives such as inclusivity personalized learning or community involvement My teaching philosophy aligns with these values and I am eager to contribute my skills passion for education and dedication to student success within your learning community These answers can be customized based on your teaching experiences educational philosophy and the specific requirements of the teaching position you are applying for Gas Journal, 1925 Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 Matter Prentice-Hall Staff.1994 Bulletin of the Atomic Scientists ,1972-10 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin's iconic Doomsday Clock stimulates solutions for a safer world Organic Chemistry, a Guided Inquiry Andrei Straumanis, 2004

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