1) Chapter 11 Liquids, Solids, and Intermolecular forces

a) Review Intermolecular Forces

1)
$$\mathbf{E} = \frac{1}{4\pi\epsilon_{*}} \frac{\mathbf{q}_{1} \mathbf{q}_{2}}{\mathbf{r}}$$

- b) Dispersion forces, Dipole-Dipole, Hydrogen Bonding, Ion-Dipole, Surface Tension, Viscosity, Capillary Action.
- c) Vaporization and Vapor Pressure, Dynamic Equilibrium
 - 1) Clausius-Clapeyron Equations

a)
$$\ln P_{\text{top}} = \frac{-\Delta H_{\text{top}}}{R} \left(\frac{1}{T}\right) + \ln \beta$$

b) $\ln \frac{P_2}{P_1} = \frac{-\Delta H_{\text{top}}}{R} \left(\frac{1}{T_2} - \frac{1}{T_1}\right)$ Two-point form

- d) Critical Point
- e) Fusion and Sublimation
- f) Phase Diagrams
- g) Intermolecular Forces and Water
- h) Band Theory (Crystalline Structures

I recommend that you review your quizzes and try a couple of problems under each section of the Exercises.

2) Chapter 12 Solutions

- a) Solutions
 - 1) Types
 - 2) Solubility of substances
 - 3) Entropy as it relates to solubility
 - 4) Effects of intermolecular forces on solubility
- b) Energetics of solution formation

a)
$$\Delta H_{soln} = \Delta H_{solne}$$
 + $\Delta H_{solvent}$ + ΔH_{mix} endothermic (+) endothermic (+) exothermic (-)

- 1) Heats of hydration (See the relationships on pages 446 & 447
- 2) Solution Equilibrium
- 3) Solution Concentration Methods
- 4) Gas as a solute (Henry's Law)
- 5 Colligative Properties
 - a) Freezing point depression and Boiling point elevation

I recommend that you review your quizzes and try a couple of problems under each section of the Exercises.

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M Woodhall

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