

Write Titrating Volume (Spreads: 1.00)

Name _____

Acetic Acid (pKa = 4.75) Titration with NaOH

a. CH_3COOH volume 25.00 mL conc. 0.100 M

b. NaOH volume 25.00 mL conc. 0.100 M

c. NaOH conc. 0.100 M compound that releases hydroxide
 NaOH conc. 0.100 M

Record Values of pH and pH for each and pH for each

pH pH pH
 pH pH pH

Species	Add NaOH , Base (pH , or both pH)	Acids with NaOH (pH or pH)	Conductivity (pH or pH)
CH_3COOH	pH	pH	pH
CH_3COO^-	pH	pH	pH
NaOH	pH	pH	pH
NaOAc	pH	pH	pH

Chemistry Note Taking Guide Episode 1101 Answers

Yijin Wang



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