Balancing & Reactions Test SCH3U

Name:

Assessment of Learning (Product) 10

Multiple Choice: Circle the most correct answer. (10K)

Which equation is improperly balanced?

A) 2 KNO₁ \rightarrow 2 KNO₂ + O₂ B) $H_{\bullet}O_{\circ} \rightarrow H_{\bullet}O + O_{\circ}$

C) $2 \text{ Na}_2O_2 + 2 \text{ H}_2O \rightarrow 4 \text{ NaOH} + O_2 D) 2 \text{H}_2O \rightarrow 2 \text{H}_2 + O_2$

In every balanced chemical equation, each side of the equation has the same number of ______.

A) atoms of each element

B) molecules

C) Modes

D) coefficients

3. Which metal would not displace lead. Pb. in a lead nitrate. Pb(NO₃b. solution?

A) copper, Cu B) magnesium, Mg C) potassium, K

D) lithium, Li E) none of the above

4. Which of the following are decomposition reactions?

 $CH_{+}(g) + O_{-}(g) \rightarrow CO_{-}(g) + H_{+}O(l)$

 $CaO(s)+CO, (g) \rightarrow CaCO, (s)$

 $Mg(s) + O_{+}(g) \rightarrow MgO(s)$

 $PbCO_{+}(s) \rightarrow PbO_{+}(s) + CO_{+}(g)$

A) 1, 2, and 3.

C3 1, 2, 3, and 4

D) 2 and 3

E) 2, 3, and 4,

Predict the product in the synthesis reaction of Al with N₁.

AN ARN

BY ALM

C) AllNo TO ALNO

E) AllNo

6. What is true of the oxide formed when magnesium reacts with oxygen?

A) les formula is Mg.O

D) It is amphoteric.

C) It is molecular

Oxides of the active metals combine with acid to form

 B) metal hydrides. A) Hi gas

B) It is basic

(C) water and a salt.

D) oxygen gas (E) metal hydroxides

8. Which one of the following compounds would produce an acidic solution when dissolved in water?

AY Na. CY B) CaO: CO Meco:

TD0 (C)O+

E) SrO:

In a double-displacement reaction, the ______.

A) products are always molecular

B) reactants are usually two ionic compounds

C) reactants are two elements

D) products are a new element and a new compound.

Chemistry 11 Chemical Reactions Test Key Answers

Shasha Hu

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