

Chemical Reactions and Equations

10

A. Magnesium and Oxygen

1. Initial appearance of Mg Shiny light gray ribbon of metal
2. Evidence of chemical reaction Heat and light energy produced
White ash produced
3. Balance: $2 \text{Mg}_{(s)} + \text{O}_{2(g)} \xrightarrow{\Delta} 2 \text{MgO}_{(s)}$
4. Type of chemical reaction: Combination reaction

B. Zinc and Copper(II) Sulfate

Time	$\text{CuSO}_4(aq)$ Appearance	$\text{Zn}(s)$ Appearance	Evidence of a Chemical Reaction
1. initial	Blue/green	Dull gray	
2. after 30 min	Lt blue	Black	Change in color

3. $\text{Zn}_{(s)} + \text{CuSO}_{4(aq)} \xrightarrow{\Delta} \text{Cu}_{(s)} + \text{ZnSO}_{4(aq)}$ (the equation is balanced)

4. Type of chemical reaction: Single replacement reaction

C. Reactions of Metals and HCl

Metal	1. Appearance of Metals	2. Evidence of a Chemical Reaction
$\text{Zn}(s)$	Gray, shiny pebbles	Formation of bubbling
$\text{Cu}(s)$	Orange, shiny strips	No change
$\text{Mg}(s)$	Light gray, shiny strips	Formation of many bubbles Heat produced

Chemical Reactions And Equations Laboratory Manual

J. A. Beran, Mark Lassiter



Chemical Reactions And Equations Laboratory Manual:

Lab Manual for General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives *Essentials of General Organic and Biochemistry* captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession To browse a sample chapter view sample ChemCasts and more visit www.whfreeman.com/gob

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Sheet to provide a space for recording observations and data during the experiment Continued emphasis on handling data is supported by the Data Analysis section General, Organic, and Biochemistry Lab Manual Sara Selfe, 2006-01-12 Offers a choice of classic chemistry experiments and innovative ones All of them place special emphasis on the biological implications of chemical concepts Available for custom publishing at <http://custompub.whfreeman.com> **Chemistry Lab Manual** Neena Sinha, R Rangarajan, R P Manchanda, R K Gupta, Rajesh Kumar, Lab Manual **List of Books for School Libraries of the State of Oregon**, 1917 **ChemDiscovery Lab Manual** Olga I. Agapova, 2002 **A Manual for the Use of the State High Schools of North Dakota** North Dakota. High School Board, 1911 **Lab Manual for Investigating Chemistry** David Collins (Ph. D.), Matthew Johll, 2008-12-02 While many of the core labs from the first edition have been retained a renewed focus on the basics of chemistry and the scientific process create an even more detailed supplemental offering

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