

Date:

#### Chemistry - Study Guide

Ch 10 Study Guide Due: \_\_\_\_\_

Chapter 10 Test -\_\_\_\_\_

When studying for this test, use your 'do nows', notes, homework, class handouts, and your textbook.

#### Vocabulary

Avogadro's number Mole molar mass equivalence statement conversion factor

For all questions involving calculations, you must show all work, round your answers to the correct number of significant figures, and label your final answers with the correct units.

#### Measuring Matter - Chapter 10 - Section 10.1 - Pgs. 320-324

- Which of the following is NOT a representative particle?
  - a. Atom
  - b. Anion
  - c. Cation
  - d. All are representative particles
- 2. Avogadro's number of representative particles is equal to one
  - a. Kilogram
  - b. Gram
  - c. Mole
  - d. Amu
- How many atoms of potassium does 1 mol of potassium contain?

4. How many moles of xenon are equivalent to 5.66 x 10<sup>26</sup> atoms?

# **Chapter 10 Study Guide Mass The Mole**

**Pamela Mills, Amina El-Ashmawy** 

#### **Chapter 10 Study Guide Mass The Mole:**

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