

Chapter 2, The Chemistry of Life (continued)**Section 2–3 Carbon Compounds** (pages 44–48)

This section explains how the element carbon is able to form millions of carbon, or organic, compounds. It also describes the four groups of organic compounds found in living things.

The Chemistry of Carbon (page 44)

1. How many valence electrons does each carbon atom have? Each carbon atom has four electrons.
2. What gives carbon the ability to form chains that are almost unlimited in length? A carbon atom can bond to other carbon atoms.

Macromolecules (page 45)

3. Many of the molecules in living cells are so large that they are known as macromolecules.
4. What is the process called by which macromolecules are formed? Polymerization
5. When monomers join together, what do they form? Polymers
6. What are four groups of organic compounds found in living things?
 - a. Carbohydrates
 - b. Lipids
 - c. Nucleic acids
 - d. Proteins

Carbohydrates (pages 45–46)

7. What atoms make up carbohydrates? Carbon, hydrogen, and oxygen atoms make up carbohydrates.
8. Circle the letter of each sentence that is true about carbohydrates.
 - ☒ a. Starches and sugars are examples of carbohydrates.
 - ☒ b. Living things use them as their main source of energy.
 - ☐ c. The monomers in sugar polymers are starch molecules.
 - ☒ d. Plants and some animals use them for strength and rigidity.
9. Single sugar molecules are also called monosaccharides.
10. Circle the letter of each monosaccharide.
☒ a. galactose b. glycogen ☒ c. glucose ☒ d. fructose
11. What are polysaccharides? They are large macromolecules formed from monosaccharides.
12. How do plants and animals store excess sugar? Plants use a polysaccharide called plant starch, whereas animals use a polysaccharide called glycogen.

Carbon Compound Study Guide Answer Key

Arshad Iqbal



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Rawn, 2014-11-04 Organic Chemistry Study Guide Key Concepts Problems and Solutions features hundreds of problems from the companion book Organic Chemistry and includes solutions for every problem Key concept summaries reinforce critical material from the primary book and enhance mastery of this complex subject Organic chemistry is a constantly evolving field that has great relevance for all scientists not just chemists For chemical engineers understanding the properties of organic molecules and how reactions occur is critically important to understanding the processes in an industrial plant For biologists and health professionals it is essential because nearly all of biochemistry springs from organic chemistry Additionally all scientists can benefit from improved critical thinking and problem solving skills that are developed from the study of organic chemistry Organic chemistry like any skill is best learned by doing It is difficult to learn by rote memorization and true understanding comes only from concentrated reading and working as many problems as possible In fact problem sets are the best way to ensure that concepts are not only well understood but can also be applied to real world problems in the work place Helps readers learn to categorize analyze and solve organic chemistry problems at all levels of difficulty Hundreds of fully worked practice problems all with solutions Key concept summaries for every chapter reinforces core content from the companion book Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry John

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