

NATURAL SCIENCES

Gr. 9 - LESSON PLAN - MATTER & MATERIALS

TOPIC: COMPOUNDS		
Sub-topic: Names of Compounds		Duration: 1 Hour
CONTENT & CONCEPTS (CAPS p63)		
KEY CONCEPTS 1. Chemical reaction 2. Chemical formula 3. Names of compounds according to their elements.		
Specific Aims:	Specific Aim 1: 'Doing Science'	X
	Specific Aim 2: 'Knowing the subject content and making connections'	X
	Specific Aim 3: 'Understanding the uses of Science'	
LESSON OBJECTIVES: 1. Learners will be able to write the names of compounds as indicated on page 63 of the CAPS Document. 2. Learners will be able to make models of different compounds		
RESOURCES REQUIRED: • Periodic Table Poster • Siyavula Sasol Inzalo (Learner book) page 163		
TEACHING & LEARNING ACTIVITY: Introduction <u>Briefly review and revise concepts dealt with in Grade 8, focusing on compounds</u> As each element has a unique name for example: - Hydrogen - Lithium Remember the first 20 elements learnt in grade 8. How did you memorize them? Do you remember Mnemonics... He - H; He - He; Lies - Li; Because - Be; Boys - B; Can - C; Not - N; Operate - O; Fireplaces - F; New - Ne; Nation - Na; Might - Mg; Also - Al; Sign - Si; Peace - P; Security - S; Clause - Cl A - Ar; King - K; Can - Ca LEARNERS MAY RECITE THIS MNEMONIC AS A WAY OF MEMORIZING THE FIRST 20 ELEMENTS. <u>The teacher provides the following explanation to the learners</u> Each compound has a unique name. How is a compound formed? - A compound is formed when TWO or more different elements combine chemically. - Each atom in a compound are held by chemical bonds - Compounds are formed during chemical reaction (when elements/ compounds combine to form new substances) - Each compound can be written as a formula, for example, H₂O; NaCl; CuSO₄ What is a formula ? - This is a notation that shows which elements the compound is made up of. - It indicates the total number of atoms in one unit of the compound - It indicates the ratio in which atoms of different elements are bonded to each other - It gives the total number of atoms in one unit of a compound. For example, the formula for Copper Sulphate is CuSO ₄		

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A Loxley



Caps Lesson Plan Grade 9 Natural Sciences:

Educational Technology and Pedagogic Encounters Yusef Waghid, 2016-07-15 This book looks at some of the underlying theories of educational technology means and ways in which this technology is guided in practice ends The authors are intent on producing ends that prepare students to undertake new analyses and evaluations that can result in new possibilities for democratic action Emphasis is on their understanding of and position within educational technology as opposed to using or applying educational technology The work is not written from the point of view that their embeddedness within educational technology has a utilitarian end in mind but rather that their situatedness within educational technology a practice in itself leaves open possibilities for new ways of understanding democratic education This book is organised into six interrelated themes that work towards the cultivation of educational technology as a human practice which guides pedagogic encounters on the basis of taking risks in relation to which the unexpected unimaginable is always possible **Annual Report** South Africa. Department of Basic Education, 2016 Primary Plans Elizabeth P. Bemis, 1911 Resources in Education, 1988 **The Popular Science Monthly**, 1921 **Popular Science**, 1972-09 Popular Science gives our readers the information and tools to improve their technology and their world The core belief that Popular Science and our readers share The future is going to be better and science and technology are the driving forces that will help make it better **Normal Instructor and Primary Plans**, 1921 Bulletin of the Atomic Scientists, 1970-12 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin's iconic Doomsday Clock stimulates solutions for a safer world Journal of Education, 1892 **Popular Science**, 1972-07

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