

## Names & Formulas for Ionic Compounds

*Give the name or formula of the following ionic compounds:*

| Name |                                            | Formula |                              |
|------|--------------------------------------------|---------|------------------------------|
| 1)   | $\text{Na}_2\text{CO}_3$ _____             | 21)     | sodium phosphide _____       |
| 2)   | $\text{NaOH}$ _____                        | 22)     | magnesium nitrate _____      |
| 3)   | $\text{MgBr}_2$ _____                      | 23)     | lead (II) sulfite _____      |
| 4)   | $\text{KCl}$ _____                         | 24)     | calcium phosphate _____      |
| 5)   | $\text{FeCl}_2$ _____                      | 25)     | ammonium sulfate _____       |
| 6)   | $\text{FeCl}_3$ _____                      | 26)     | silver cyanide _____         |
| 7)   | $\text{Zn}(\text{OH})_2$ _____             | 27)     | aluminum sulfide _____       |
| 8)   | $\text{Be}_2\text{SO}_4$ _____             | 28)     | beryllium chloride _____     |
| 9)   | $\text{CrF}_2$ _____                       | 29)     | copper (I) arsenide _____    |
| 10)  | $\text{Al}_2\text{S}_3$ _____              | 30)     | iron (III) oxide _____       |
| 11)  | $\text{PbO}$ _____                         | 31)     | gallium nitride _____        |
| 12)  | $\text{Li}_3\text{PO}_4$ _____             | 32)     | iron (II) bromide _____      |
| 13)  | $\text{TiI}_4$ _____                       | 33)     | vanadium (V) phosphate _____ |
| 14)  | $\text{Co}_3\text{N}_2$ _____              | 34)     | calcium oxide _____          |
| 15)  | $\text{Mg}_3\text{P}_2$ _____              | 35)     | magnesium acetate _____      |
| 16)  | $\text{Ga}(\text{NO}_2)_3$ _____           | 36)     | aluminum sulfate _____       |
| 17)  | $\text{Ag}_2\text{SO}_3$ _____             | 37)     | copper (I) carbonate _____   |
| 18)  | $\text{NH}_4\text{OH}$ _____               | 38)     | barium oxide _____           |
| 19)  | $\text{Al}(\text{CN})_3$ _____             | 39)     | ammonium sulfite _____       |
| 20)  | $\text{Be}(\text{CH}_3\text{COO})_2$ _____ | 40)     | silver bromide _____         |

# Compound Names And Formulas Activity Answer Key

**Tina Overton, Jonathan Rourke, Fraser  
A. Armstrong**



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