

ACIDS and BASES: Multiple Choice

1. Apples have a pH of 3.9, beans have a pH of 5.9, and coconuts a pH of 7.9.

Which of the following statements is true?

- A) Beans are 100 times more acidic than coconut, and 10 000 times more acidic than apples.
B) Apples are 100 times more acidic than beans, and 10 000 times more acidic than coconut.
C) Beans are 100 times more acidic than apples, and 10 000 times more acidic than coconut.
D) Coconut is 100 times more acidic than beans, and 10 000 times more acidic than apples.

ANS: B

1. Which of the following is NOT a strong base?

- A) $\text{NaOH}_{(aq)}$ B) $\text{KOH}_{(aq)}$ C) $\text{LiOH}_{(aq)}$ D) $\text{CaOH}_{(aq)}$ E) $\text{Mg(OH)}_{2(aq)}$

ANS: D

1. Which compound is a strong base?

- A) B(OH)_3 B) KOH C) SO_4^{2-} D) HCO_3^- E) NH_3

ANS: B

1. Which of the following indicates an acidic solution?

- A) pH = 7.5 B) pH = 7 C) pH = 6.5
D) more than one correct response E) no correct response

ANS: C

1. Which of the following indicates an acidic solution?

- A) pH = 5.5 B) pH = 7 C) pH = 6.5
D) more than one correct response E) no correct response

ANS: D

1. Which of the following indicates an acidic solution?

- A) pH = 8.5 B) pH = 7 C) pH = 7.5
D) more than one correct response E) no correct response

ANS: E

1. Which of the following equations represents an acid-base neutralization reaction?

- A) $\text{H}_2\text{SO}_4 + \text{Zn} \rightarrow \text{ZnSO}_4 + \text{H}_2$ B) $\text{HNO}_3 + \text{KOH} \rightarrow \text{KNO}_3 + \text{H}_2\text{O}$
C) $\text{Ba(OH)}_2 + \text{Na}_2\text{SO}_4 \rightarrow \text{BaSO}_4 + 2\text{NaOH}$ D) more than one correct response
E) none of the above

ANS: B

Chemistry Test Acids Bases Multiple Choice

JR Anderson



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