

Section 14.3 Solvation and Solubility?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible
insoluble

liquid
miscible

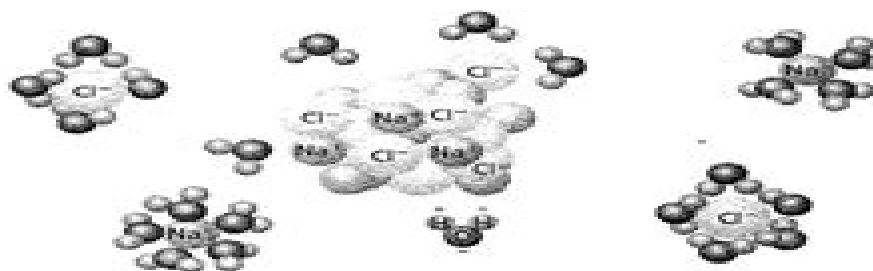
soluble
solute

solution
solvent

Air is a(n) **(1)** _____ of oxygen gas dissolved in nitrogen gas. The oxygen in air is the **(2)** _____, and nitrogen is the **(3)** _____. Because oxygen gas dissolves in a solvent, oxygen gas is a(n) **(4)** _____ substance. A substance that does not dissolve is **(5)** _____. **(6)** _____ solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are **(7)** _____. However, if one liquid is insoluble in another, the liquids are **(8)** _____.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.

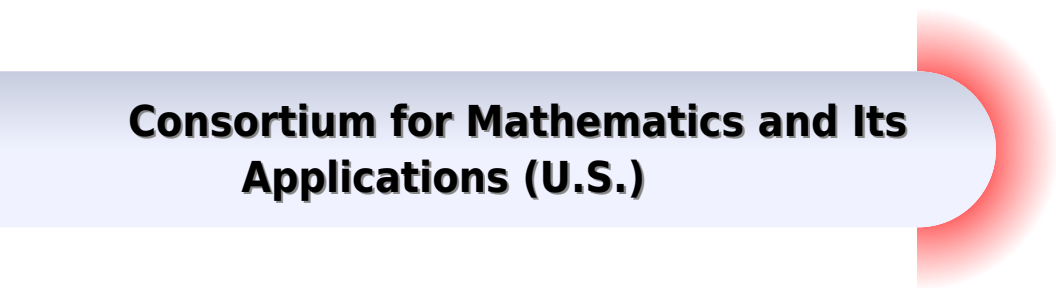


9. Hydration is solvation in which the solvent is water. What is solvation?

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Chapter 6

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