

Chemistry
Balancing Equations

Name Answer Key

Date _____ Per _____ Row _____

1. 2 Fe + 3 H₂SO₄ → Fe₂(SO₄)₃ + 3 H₂
2. 2 C₂H₆ + 7 O₂ → 6 H₂O + 4 CO₂
3. 3 KOH + H₃PO₄ → K₃PO₄ + 3 H₂O
4. SnO₂ + 2 H₂ → Sn + 2 H₂O
5. 4 NH₃ + 5 O₂ → 4 NO + 6 H₂O
6. 2 KNO₃ + H₂CO₃ → K₂CO₃ + 2 HNO₃
7. B₂Br₆ + 6 HNO₃ → 2 B(NO₃)₃ + 6 HBr
8. 2 BF₃ + 3 Li₂SO₃ → B₂(SO₃)₃ + 6 LiF
9. 4 (NH₄)₃PO₄ + 3 Pb(NO₃)₄ → Pb₃(PO₄)₄ + 12 NH₄NO₃
10. SeCl₆ + O₂ → SeO₂ + 3 Cl₂
11. CH₄ + 2 O₂ → CO₂ + 2 H₂O
12. Na⁺ + Cl⁻ → NaCl
13. 4 Al + 3 O₂ → 2 Al₂O₃
14. N₂ + 3 H₂ → 2 NH₃
15. 8 CO(g) + 17 H₂(g) → C₈H₁₈(l) + 8 H₂O
16. FeO₃(s) + 3 CO(g) → Fe(l) + 3 CO₂(g)
17. 2 H₂SO₄ + Pb(OH)₄ → Pb(SO₄)₂ + 4 H₂O
18. 2 Al + 6 HCl → 2 AlCl₃ + 3 H₂
19. Ca₃(PO₄)₂ + 2 H₂SO₄ → 2 CaSO₄ + Ca(H₂PO₄)₂
20. H₃PO₄ + 5 HCl → PCl₅ + 4 H₂O

Data Table 1 Chemical Equations Key

Puja Mehta

A decorative red circular graphic with a gradient, appearing as a partial circle on the right side of the slide.

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and understanding of the field and the role it will play in their fields of further study **Science Vocabulary Building, Grades 5 - 8** Schyrlet Cameron, Carolyn Craig, 2009-02-16 Connect students in grades 5 8 with science using Science Vocabulary Building This 80 page book reinforces commonly used science words builds science vocabulary and increases students readability levels This comprehensive classroom supplement includes alphabetized word lists that provide pronunciations syllabifications definitions and context sentences for high utility science words Activities allow for differentiated instruction and can be used as warm ups homework assignments and extra practice The book supports National Science Education Standards

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