

Chemistry EOC Review Packet

Matter and Change: Pages 34-50

Procedural:

- Chemical change
- Chemical property
- Physical change
- Physical property
- Heterogeneous mixture
- Heterogeneous mixture
- Compound
- Element
- Gas
- Liquid
- Solid
- Law of Conservation of Mass

Concept Questions:

- What is the definition of matter? Give 3 examples of substances and 3 that are not substances.

Something that has mass and takes up space: (matter) iron, wood, water, hydrogen gas, steel

- Fill in the chart with whether matter should be classified as solid, liquid, or gas.

States of Matter	Solid	Liquid
Shape	fixed	variable
Volume	fixed	variable
Mass	fixed	variable

- What are the properties of each state of matter?



Solid



Liquid



Gas

- What properties distinguish plasmas?

Electrically charged particles, ionized, superheated

- What properties distinguish supercritical?

Chemistry Matter And Change Review Packet Answers

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