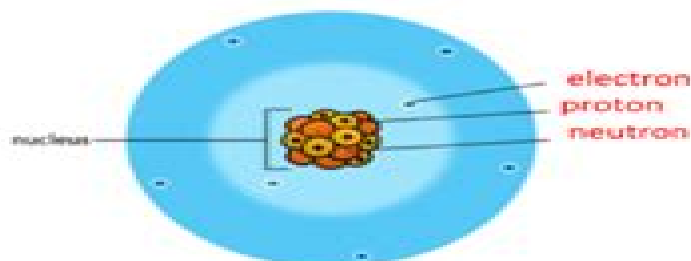


CHEMISTRY TEST STUDY GUIDE ANSWER KEY

Complete the following information about atoms.
Label the subatomic particles in the atom diagram below.



What **charge** does each subatomic particle have?

Proton: **positive**

Electron: **negative**

Neutron: **neutral (no charge)**

What subatomic particles are found in the **nucleus**?

protons and neutrons

Classify each of the following substances as an Element (E), Compound (C), or Mixture (M).

If it's a mixture, is it heterogeneous (he) or homogeneous (ho)?

- water **C**
- salt **C**
- air **M (ho)**
- carbon **E**
- sugar **C**
- saltwater **M (ho)**
- hydrogen **E**
- carbon dioxide **C**
- kool aid **M (ho)**

Use a periodic table to answer the following questions.

- Is sodium a metal or nonmetal? **metal**
- What is the atomic number of oxygen? **8**
- What is element symbol for iron? **Fe**
- What other elements are in the same family as Chlorine? **Fluorine, Bromine, Iodine, Astatine, Tennessine**
- Li, Be, B, C, N, O, F, and Ne are all in the same: **Period**

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Charles H. Corwin



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Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text

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